

Project Notes:

Project Title: Filtering Microplastic

Name: Neena Xiang

Note Well: There are NO SHORT-cuts to reading journal articles and taking notes from them. Comprehension is paramount. You will most likely need to read it several times so set aside enough time in your schedule.

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Knowledge Gaps:

This list provides a brief overview of the major knowledge gaps for this project, how they were resolved and where to find the information.

Knowledge Gap	Resolved By	Information is located	Date resolved
What is the composition of microplastics			
How can microplastic be detected?	Reading article	Notes in Article 2	October 16, 2019
What are the most common types of plastics?	Reading article	Notes in Article 2	October 16, 2019
Properties of common plastic			

Literature Search Parameters:

These searches were performed between (Start Date of reading) and XX/XX/2019.

List of keywords and databases used during this project.

Database/search engine	Keywords	Summary of search
Engineering Village	Microplastic and filter	Found filters used on seawater
Engineering Village	Microplastic and filter	Found ways to detect microplastic
ScienceDirect	Common plastics and properties	Found a chapter of a book that explained the properties of common plastics

Article #1 Notes: Title

Article notes should be on separate sheets

Source Title	
Source Author	
Source citation	
Original URL	
Source type	
Keywords	
Summary of key points	
Important Figures	
Reason for interest	
Notes	
Follow up Questions	

Article #1 Notes: Microplastic pollution in deep-sea sediments

Article notes should be on separate sheets

Source Title	Environmental Science
Source Author	Lisbeth Van Cauwenberghe
Source citation	Van Cauwenberghe, L., Vanreusel, A., Mees, J., & Janssen, C. R. (2013). Microplastic pollution in deep-sea sediments. <i>Environmental Pollution</i> . Retrieved from http://onemoregeneration.org/wp-content/uploads/2012/07/Microplastic-pollution-in-deep-sea-sediments.pdf
Original URL	http://onemoregeneration.org/wp-content/uploads/2012/07/Microplastic-pollution-in-deep-sea-sediments.pdf
Source type	Online journal
Keywords	Microplastics, Deep sea, Sediment, Pollution
Summary of key points	<ul style="list-style-type: none"> - Microplastics are found in deeper depths than initially thought - Samples were taken from deeper in the ocean and microplastics were found

Important Figures

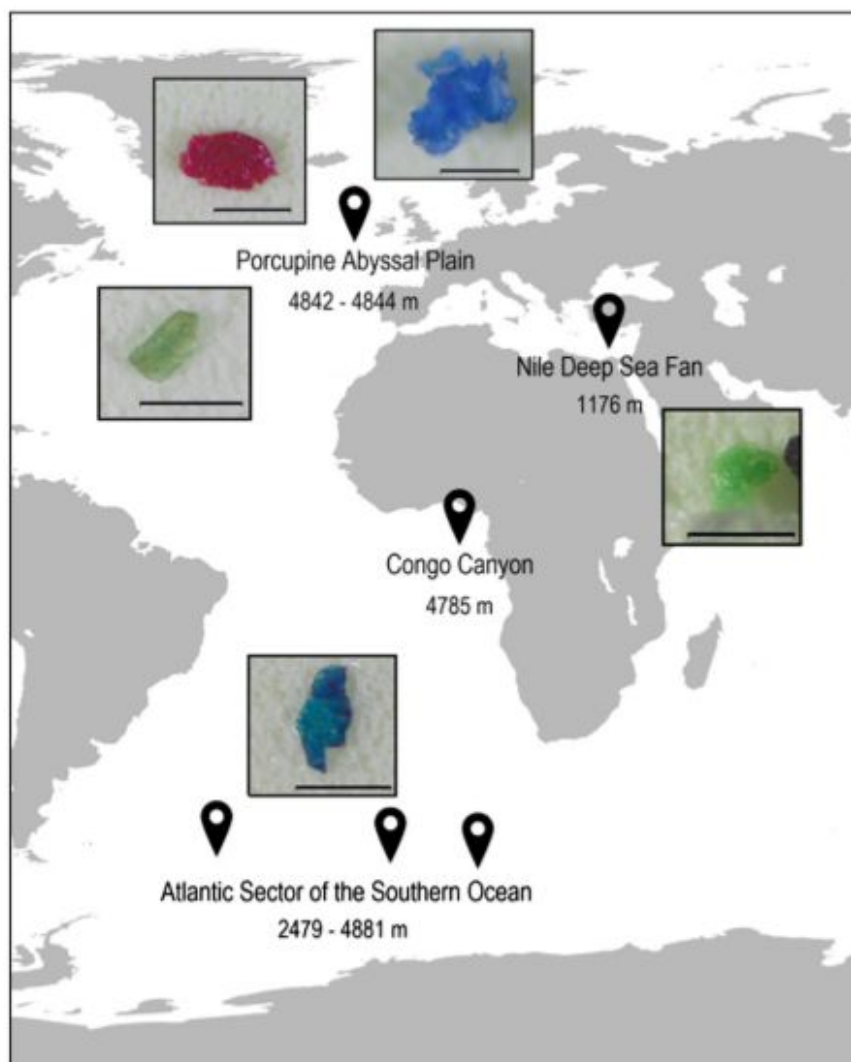


Fig. 1. Locations of the sites sampled and microplastics recovered from the deep sea sediments. Three locations in the Atlantic Ocean (Porcupine Abyssal Plain, Lobe of Congo Canyon and Atlantic sector of the Southern ocean) and one location in the Mediterranean Sea (Nile Deep Sea Fan) were sampled. Microplastics (see inserted photos) were detected in the top centimetre of sediments from three of the four sampled locations (Scale bar represents 100 μm).

L. Van Cauwenberghe et al. / Environmental Pollution 182 (2013) 495–499

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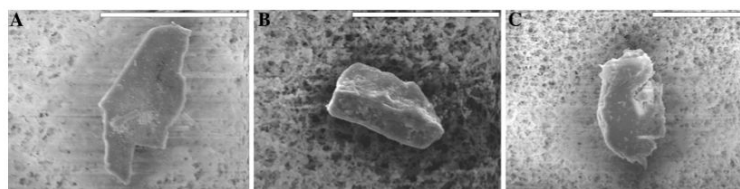


Fig. 2. Scanning electron microscopy images of microplastics. A. Blue particle extracted from sediment from the Southern ocean (2749 m). B. Green particle originating from the Porcupine Abyssal Plain (4842 m). C. Red particle extracted from sediment from the Porcupine Abyssal Plain (4843 m) (Scale bar represent 100 μm).

Reason for interest

There may be microplastic found in more places than the surface waters of the ocean.

Notes	<ul style="list-style-type: none"> - Improper plastic disposal leads to the breakdown of those plastics in the ocean - Sewage treatments don't keep microplastic, so it is released into aquatic environments - " plastic particles <1mm accounted for 65% of all marine debris items collected on beaches in the Tamar Estuary (U.K.)" - Marine animals will eat the plastics, and it will either exit the animal as defecation or stay in the tissues of the animal. - There is plastic found in water columns and sediment in the ocean
Follow up Questions	Could there be micro plastic in the sediment on land?

Microplastics are small particles of plastic which have broken down from bigger plastic items. There has been a build up for decades, but it wasn't until recently that people have wondered how far the pollution has gotten. There have been microplastics found on the surface of the ocean, but there may be more microplastics at the bottom of the ocean. The study was conducted to prove that microplastic has indeed permeated to the bottom of the ocean.

Article #2 Notes: Occurrence, identification and removal of microplastic particles and fibers in conventional activated sludge process and advanced MBR technology

Article notes should be on separate sheets

Source Title	Water Research
Source Author	Mirka Lares
Source citation	Lares, M., Ncibi, M. C., & Sillanpaa, M. (2018). Occurrence, identification and removal of microplastic particles and fibers in conventional activated sludge process and advanced MBR technology. <i>Water Research</i> , 133, 236–246. doi: https://doi.org/10.1016/j.watres.2018.01.049
Original URL	https://www.sciencedirect.com/science/article/pii/S0043135418300630
Source type	Online Journal
Keywords	Microplastics, WWTP (wastewater treatment plants), Wastewater, Sludge, Identification
Summary of key points	<ul style="list-style-type: none"> - Sludge samples were collected for 3 months - The two methods of filtration used were conventional activated sludge and membrane bioreactor. - Activated sludge was more efficient.

Important Figures

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M. Lares et al. / Water Research 133 (2018) 236–246

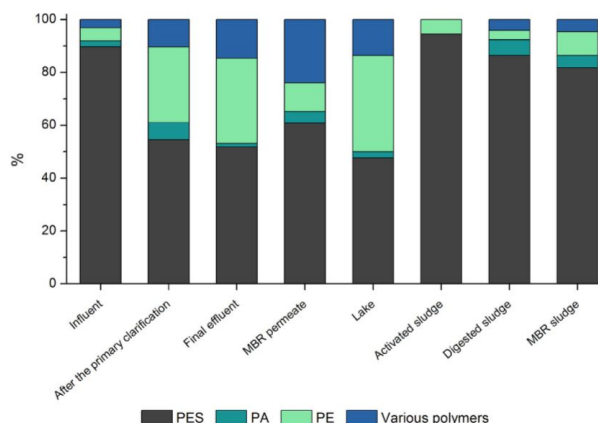


Fig. 1. The proportion of different polymers in different stages of the WWTP and a nearby lake. Various polymers include e.g. PE, PP and PES fragments with similar kind of appearance.

M. Lares et al. / Water Research 133 (2018) 236–246

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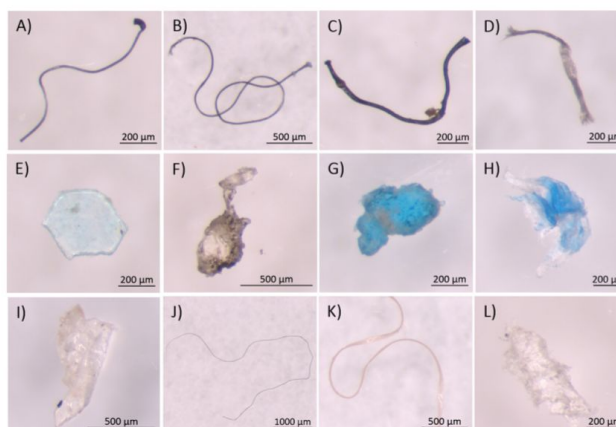


Fig. 2. Typical appearance of different polymers detected in different stages of the WWTP and recipient lake and identified by micro-FTIR and/or micro-Raman. (A–E) Polyester, (F–I) polyethylene, (J–K) polyamide and (L) polypropylene.

Reason for interest

This article presented some methods of filtration of microplastics

Notes

- Plastic gets into the environment through the disposal of plastic products and the wear and tear of other frequently used products.
- There are fluctuations between studies of the size of microplastic
- Most particles are removed from wastewater and end up in treatment plants
- Sludge may end up in the environment which also releases the microplastic in it
- There may be even smaller nanoplastics
- Possible means of filtration include infrared spectroscopy, Raman spectroscopy, and visually looking at it
- Great caution was taken to avoid contamination
- Polyester, polyethylene, polyamide, and polypropylene are common plastics

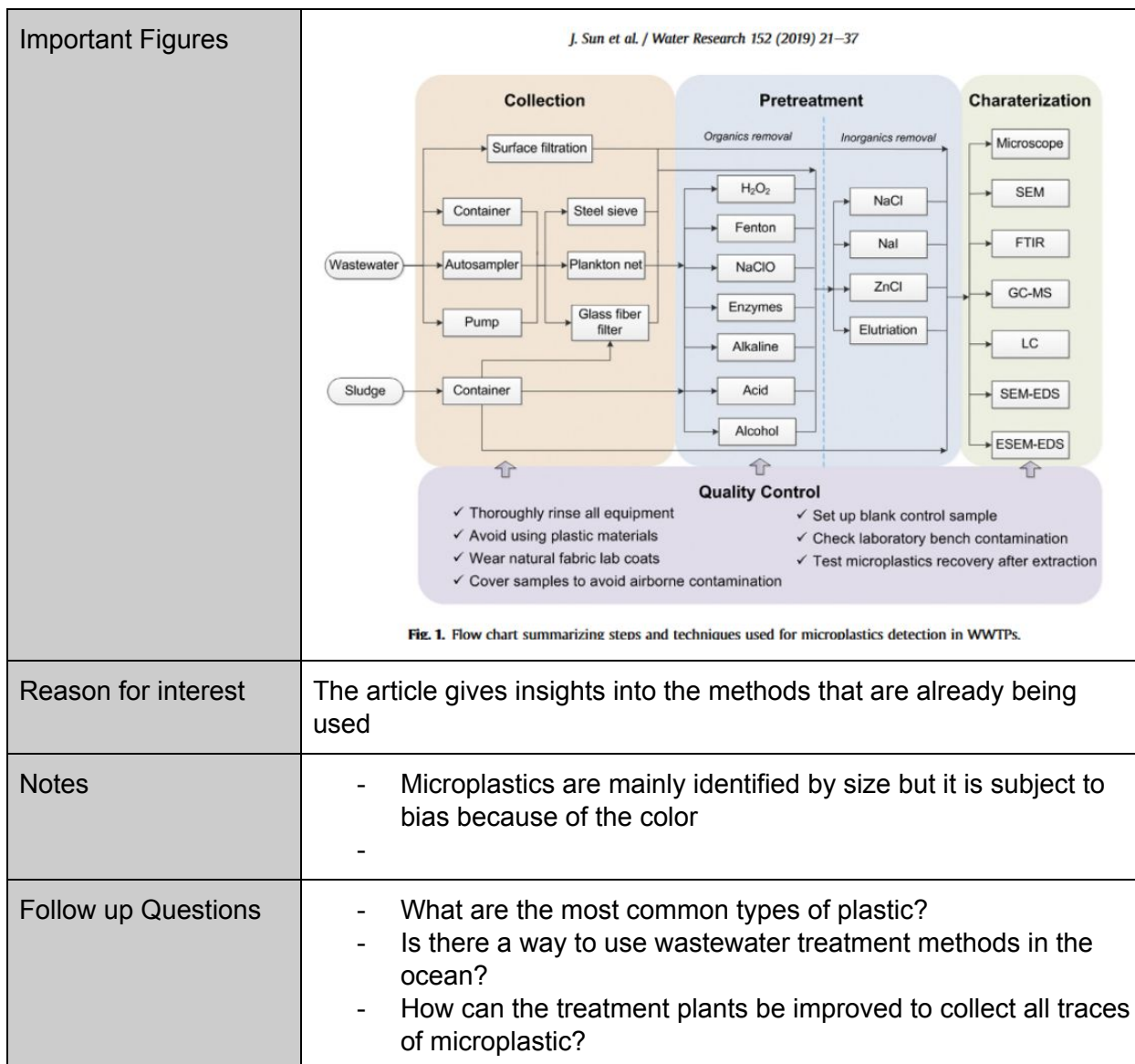
	<ul style="list-style-type: none"> - Polyester fibers were the most common plastics in the samples - Polyethylene was the most common polymer in the samples - Cosmetics give off microplastics - Microplastic fibers are more common than microplastic particles in wastewater sludge - If a mesh smaller than 250 nanometers is used, small sized MPs are lost
Follow up Questions	<p>What other methods are there to filter out microplastic?</p> <p>What problems do different types of plastics create when being collected?</p> <p>How many different types of plastics are there?</p>

Microplastic has found its way to wastewater treatment plants. The goal of the study was to find a way to efficiently filter the microplastic from the collected sewage. Researchers used the conventional activated sludge and membrane bioreactor to filter the microplastic, and identified any pieces with a microscope. The conventional activated sludge was found to be more efficient, and the collected microplastics were released into bodies of water.

Article #3 Notes: Microplastics in wastewater treatment plants: Detection, occurrence and removal

Article notes should be on separate sheets

Source Title	Water Research
Source Author	Jing Sun
Source citation	Sun, J., Dai, X., Wang, Q., van Loosdrecht, M. C. M., & Ni, B.-jie. (n.d.). Microplastics in wastewater treatment plants: Detection, occurrence and removal. <i>Water Research</i> , 152, 21–37. Retrieved from Microplastics in wastewater treatment plants: Detection, occurrence and removal
Original URL	https://www.sciencedirect.com/science/article/pii/S0043135418310686
Source type	Online Journal
Keywords	Microplastics Wastewater treatment plant detection Occurrence Sewage sludge Treatment technologies
Summary of key points	<ul style="list-style-type: none"> - There is no immediate danger in consuming microplastic, but the toxins can build up -



Article #4 Notes: Small Microplastic Sampling in Water: Development of an Encapsulated Filtration Device

Article notes should be on separate sheets

Source Title	Small Microplastic Sampling in Water: Development of an Encapsulated Filtration Device
Source Author	Robin Lenz and Matthias Labrenz
Source citation	Lenz, R., & Labrenz, M. (2018). Small Microplastic Sampling in Water: Development of an Encapsulated Filtration Device. <i>Water</i> , 10(8). doi: https://doi.org/10.3390/w10081055
Original URL	https://www.mdpi.com/2073-4441/10/8/1055/htm
Source type	Journal Article
Keywords	Microplastics, field work, pollution monitoring, engineering design
Summary of key points	- Microplastics that are less than 10 nanometers become too hard for flow through filtration to work
Important Figures	<p>The diagram illustrates the development process of an encapsulated filtration device, divided into five stages:</p> <ul style="list-style-type: none"> User Requirements: <ul style="list-style-type: none"> sample MP from surface and subsurface waters target small MP ($\geq 10 \mu\text{m}$) easy mobility; versatile field application minimise contamination risk System Requirements: <ul style="list-style-type: none"> equipped with interchangeable sample intake solutions retention of $10 \mu\text{m}$ particles possible rugged, self-contained, rollable, modular setup system-caused contamination or MP destruction excluded by design Components Specification: <ul style="list-style-type: none"> μ-filtration on exchangeable stainless steel meshes of $10 \mu\text{m}$ pore size plastic free design (except PTFE) suction instead of pressurised pumping extendable cartridge filter units and off the shelf fittings Design Validation: <ul style="list-style-type: none"> first lab tests confirmed principal functioning: reverse flushing for contamination prevention, suction sampling and flow measurement field tests showed achievable sample volumes on $4 \times 5''$ filters recovery rates tested by spiking experiments Field Application + Improvement: <ul style="list-style-type: none"> additional 4-way valve for in-process change of flow direction: faster and easier start increase of maximum filter capacity usage future possibilities: <ul style="list-style-type: none"> cascade setup increase of filtration capacity
Reason for interest	This journal talk about some of the filtration methods that area

	already being used
Notes	<ul style="list-style-type: none">- Most filtration methods just skim the surface of the water and need water to flow through it to work- Centrifuges have also been used to separate microplastics by density- When trying to collect really small microplastics, the filter material and air will contaminate the sample.- Really fine sieves are easily clogged-
Follow up Questions	<ul style="list-style-type: none">- How can I get around flow through filtration?- Is there a need for me to attempt to filter these smaller sized plastics?

Article #5 Notes: Size and shape matter: A preliminary analysis of microplastic sampling technique in seawater studies with implications for ecological risk assessment

Article notes should be on separate sheets

Source Title	Size and shape matter: A preliminary analysis of microplastic sampling technique in seawater studies with implications for ecological risk assessment
Source Author	Garth A Coverton, Christopher M Pearce, Helen J Gurney-Smith, Stephen G Chastain, Peter S. Ross, John F. Dower, Sarah E Dudas
Source citation	Covernton, G. A., Pearce, C. M., Gurney-Smith, H. J., Chastain, S. G., Ross, P. S., Dower, J. F., & Dudas, S. E. (2019). Size and shape matter: A preliminary analysis of microplastic sampling technique in seawater studies with implications for ecological risk assessment. <i>Science of the Total Environment</i> , 667, 124–132. Retrieved from https://www-sciencedirect-com.ezproxy.wpi.edu/science/article/pii/S004896971930854X
Original URL	https://www-sciencedirect-com.ezproxy.wpi.edu/science/article/pii/S004896971930854X
Source type	Journal
Keywords	Aquaculture, Discrete, Filtration, Mesh, Methodology, Microplastics
Summary of key points	<ul style="list-style-type: none"> - Microfibers are the most common shape of microplastic - They can pass through mesh due to their thickness - Mesh may not be the best method of filtration since it measures 1 - 4 times less

Important Figures

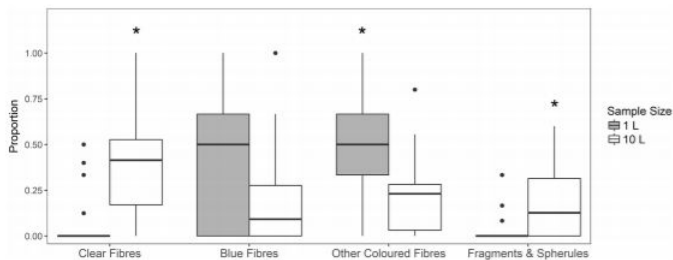
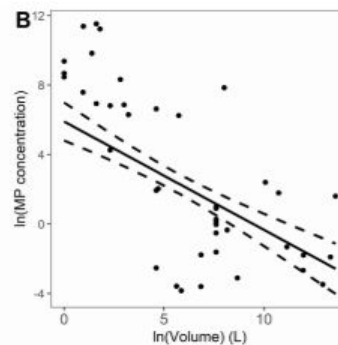
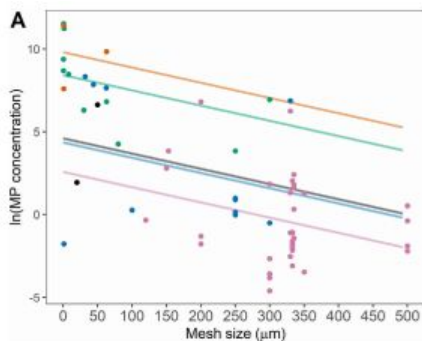
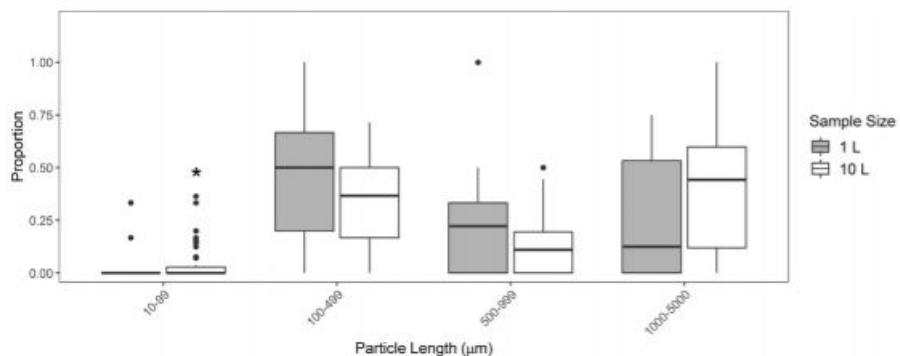


Fig. 3. Boxplots of proportions of potential microplastic particle types in 36 samples, separated by sampling method (1-L jar sample and 10-L filtered bucket sampling). Asterisks indicate significant differences between methods as determined by two-sample Mann-Whitney *U* tests. Coloured fibres (aside from blue) are grouped together, as well as fragments and spherules. Box limits and whiskers show first and third quartiles, respectively, with outliers (>1.5 times the distance between the first and third quartiles away from the median) shown as points.

G.A. Coverton et al. / Science of the Total Environment 667 (2019) 124–132



Reason for interest

- Talks about sampling methods and possible environmental risks
- Shows the types of particles that were in samples

Notes

- Larger meshes underestimate the amount of microplastic found in water

Follow up Questions

How can I implement other filtration methods that aren't mesh?

Article #6 Notes: Development of an optimal filter substrate for the identification of small microplastic particles in food by micro-Raman spectroscopy

Article notes should be on separate sheets

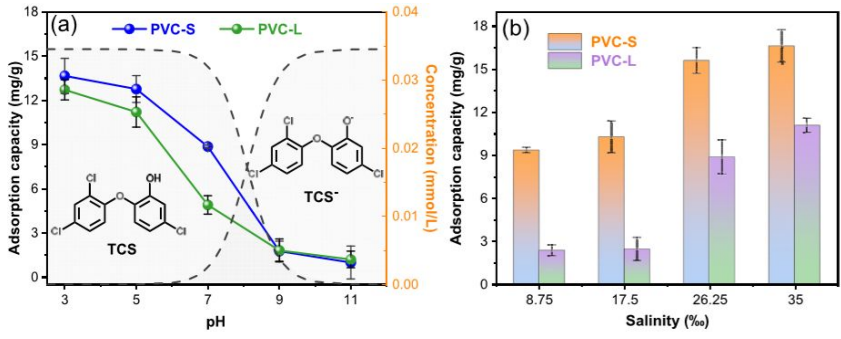
Source Title	Development of an optimal filter substrate for the identification of small microplastic particles in food by micro-Raman spectroscopy																																																																															
Source Author	Barbara E Obmann, George Sarau, Sebastian W Schmitt, Heinrich Holtmannspötter, Silke H Christiansen Wilhelm Dicke																																																																															
Source citation	Oßmann, B.E., Sarau, G., Schmitt, S.W. et al. Anal Bioanal Chem (2017) 409: 4099. https://doi-org.ezproxy.wpi.edu/10.1007/s00216-017-0358-y																																																																															
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Keywords	Microplastics . Food . Micro-Raman spectroscopy . Aluminium . Polycarbonate membrane filter . Substrate																																																																															
Summary of key points	<ul style="list-style-type: none"> - Analysis of filtration usually happens on surface of filter - Membrane filters are helpful filters 																																																																															
Important Figures	<p>Table 1 Material, manufacturer, pore size and diameter of all tested, commercially available membrane filters</p> <table border="1"> <thead> <tr> <th>Material</th> <th>Manufacturer</th> <th>Pore size (µm)</th> <th>Diameter (mm)</th> </tr> </thead> <tbody> <tr> <td>Regenerated cellulose</td> <td>GE Healthcare Life Sciences Whatman™</td> <td>0.45</td> <td>50</td> </tr> <tr> <td>Nitrocellulose</td> <td>Macherey-Nagel</td> <td>0.45</td> <td>47</td> </tr> <tr> <td>Cellulose acetate</td> <td>Macherey-Nagel</td> <td>0.45</td> <td>47</td> </tr> <tr> <td>Aluminium oxide</td> <td>GE Healthcare Life Sciences Whatman™</td> <td>0.2</td> <td>25/47</td> </tr> <tr> <td>Silver</td> <td>Pieper Filter GmbH</td> <td>0.45</td> <td>47</td> </tr> <tr> <td>PC white</td> <td>Merck Millipore Ltd.</td> <td>0.4</td> <td>25</td> </tr> <tr> <td>PC black</td> <td>Merck Millipore Ltd.</td> <td>0.4</td> <td>25</td> </tr> <tr> <td rowspan="2">PC coated with gold</td> <td>Nuclepore GmbH Costar</td> <td>0.8</td> <td>37</td> </tr> <tr> <td>GE Healthcare Life Sciences Whatman™</td> <td>0.4</td> <td>25</td> </tr> <tr> <td></td> <td></td> <td>0.8</td> <td>13</td> </tr> <tr> <td></td> <td>APC</td> <td>0.4</td> <td>25</td> </tr> <tr> <td></td> <td></td> <td>0.8</td> <td>25</td> </tr> </tbody> </table> <p>Table 2 Material, type, manufacturer and size of all used plastic particle standards</p> <table border="1"> <thead> <tr> <th>Material</th> <th>Type</th> <th>Manufacturer</th> <th>Size (µm)</th> </tr> </thead> <tbody> <tr> <td>Polystyrene</td> <td>Polybead micron microspheres, 2.5% solids in water</td> <td>Polysciences Inc.</td> <td>1</td> </tr> <tr> <td rowspan="2">Polyethylene</td> <td rowspan="2">Clear microspheres, powder</td> <td rowspan="2">Cospheric</td> <td>1–10</td> </tr> <tr> <td>10–106</td> </tr> <tr> <td>Polypropylene</td> <td>Chromatographic grade, powder</td> <td>Polysciences Inc.</td> <td>25–85</td> </tr> <tr> <td>Polyamide-Nylon 6</td> <td>Powder</td> <td>GoodFellow</td> <td>15–20 (average particle size)</td> </tr> <tr> <td>Poly(vinyl chloride)</td> <td>Powder</td> <td>Pyropowders.de</td> <td><50</td> </tr> </tbody> </table>				Material	Manufacturer	Pore size (µm)	Diameter (mm)	Regenerated cellulose	GE Healthcare Life Sciences Whatman™	0.45	50	Nitrocellulose	Macherey-Nagel	0.45	47	Cellulose acetate	Macherey-Nagel	0.45	47	Aluminium oxide	GE Healthcare Life Sciences Whatman™	0.2	25/47	Silver	Pieper Filter GmbH	0.45	47	PC white	Merck Millipore Ltd.	0.4	25	PC black	Merck Millipore Ltd.	0.4	25	PC coated with gold	Nuclepore GmbH Costar	0.8	37	GE Healthcare Life Sciences Whatman™	0.4	25			0.8	13		APC	0.4	25			0.8	25	Material	Type	Manufacturer	Size (µm)	Polystyrene	Polybead micron microspheres, 2.5% solids in water	Polysciences Inc.	1	Polyethylene	Clear microspheres, powder	Cospheric	1–10	10–106	Polypropylene	Chromatographic grade, powder	Polysciences Inc.	25–85	Polyamide-Nylon 6	Powder	GoodFellow	15–20 (average particle size)	Poly(vinyl chloride)	Powder	Pyropowders.de	<50
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Reason for interest	It explores the concept of micro-Raman spectroscopy
Notes	<ul style="list-style-type: none"> - There is microplastic even in food. - Microplastics can get into organs through food consumption - Using a metal coating will help material be detected by Raman spectroscopy - Regenerated cellulose, nitrocellulose and cellulose acetate are some good materials.
Follow up Questions	Are there other more effective methods of analyzing collected plastic?

Article #7 Notes: Effect of microplastic size on the adsorption behavior and mechanism of triclosan on polyvinyl chloride

Article notes should be on separate sheets

Source Title	Effect of microplastic size on the adsorption behavior and mechanism of triclosan on polyvinyl chloride
Source Author	Jie Ma, Jinghua Zhao, Zhilin Zhu, Liquing Li, Yu Fei
Source citation	<p>Ma, J., Zhao, J., Zhu, Z., Li, L., & Yu, F. (2019). Effect of microplastic size on the adsorption behavior and mechanism of triclosan on polyvinyl chloride. <i>Environmental Pollution</i>, 254</p> <p>Retrieved from https://www.sciencedirect.com/science/article/pii/S0269749119336310</p>
Original URL	https://www.sciencedirect.com/science/article/pii/S0269749119336310
Source type	Journal article
Keywords	Microplastics, Polyvinyl chloride, Triclosan, Size, Adsorption behavior

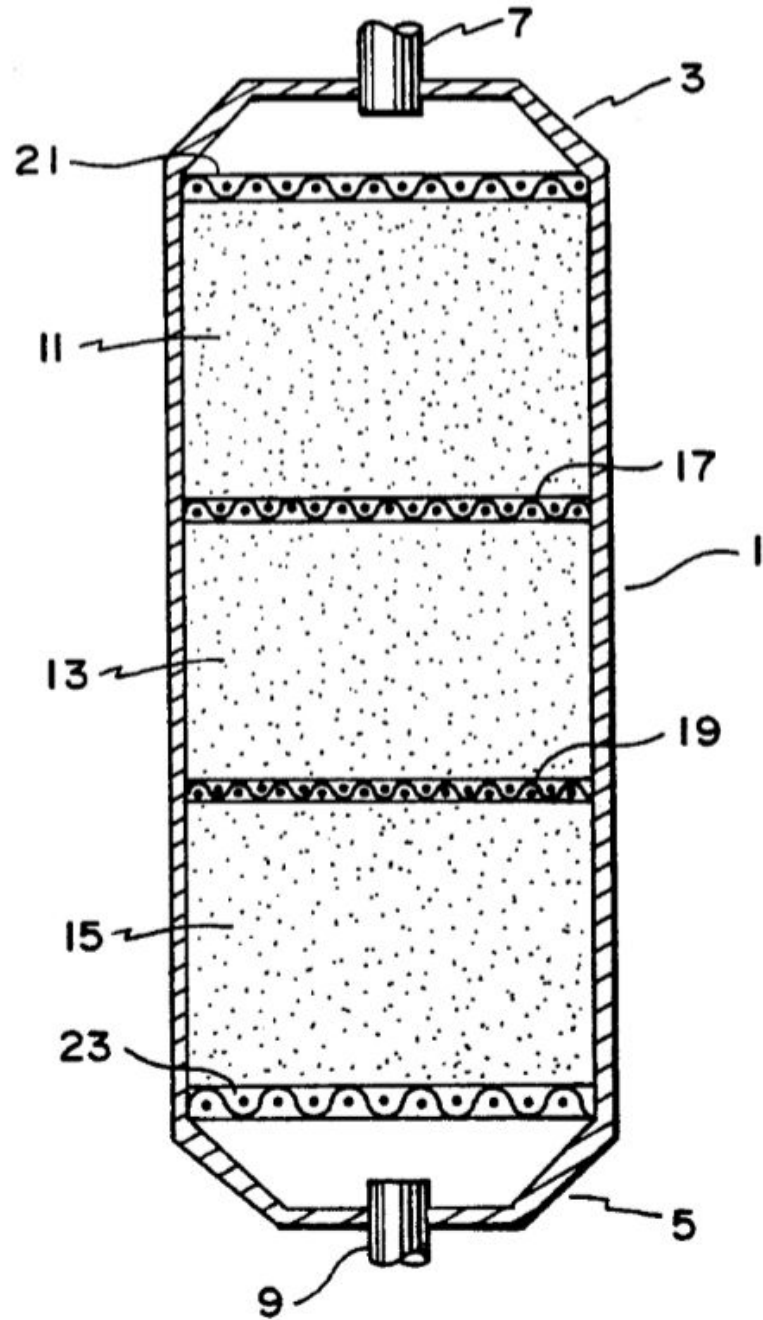
Summary of key points	<ul style="list-style-type: none"> - Particle sizes affected absorption capacity - Salinity and pH also changed the amount of pollutant that could be absorbed. 																																																																																									
Important Figures	 <p>Fig. 4. The influences of pH (a) and salinity (b) on adsorption of TCS on PVC-S and PVC-L.</p> <p>Table 2 Adsorption isotherm parameters of TCS on PVC-S and PVC-L (PVC: 10 mg, T = 298 K).</p> <table border="1" data-bbox="535 787 1404 1291"> <thead> <tr> <th rowspan="2"></th> <th colspan="3">Linear</th> <th colspan="4">Langmuir</th> </tr> <tr> <th>k_d</th> <th>R^2</th> <th>P</th> <th>q_e</th> <th>k_L</th> <th>R^2</th> <th>P</th> </tr> </thead> <tbody> <tr> <td>PVC-S</td> <td>1.35</td> <td>0.856</td> <td>0.0028</td> <td>42.1</td> <td>0.061</td> <td>0.915</td> <td>6.00E-4</td> </tr> <tr> <td>PVC-L</td> <td>1.05</td> <td>0.865</td> <td>0.0024</td> <td>40.2</td> <td>0.044</td> <td>0.909</td> <td>7.52E-4</td> </tr> </tbody> </table> <table border="1" data-bbox="535 955 1404 1123"> <thead> <tr> <th rowspan="2"></th> <th colspan="4">Freundlich</th> <th colspan="4">Temkin</th> </tr> <tr> <th>k_F</th> <th>n</th> <th>R^2</th> <th>P</th> <th>B_T</th> <th>k_T</th> <th>R^2</th> <th>P</th> </tr> </thead> <tbody> <tr> <td>PVC-S</td> <td>3.01</td> <td>1.38</td> <td>0.892</td> <td>1.17E-4</td> <td>6.77</td> <td>0.974</td> <td>0.878</td> <td>1.59E-4</td> </tr> <tr> <td>PVC-L</td> <td>2.05</td> <td>1.30</td> <td>0.890</td> <td>1.34E-4</td> <td>5.77</td> <td>0.803</td> <td>0.866</td> <td>2.19E-4</td> </tr> </tbody> </table> <table border="1" data-bbox="535 1123 1404 1291"> <thead> <tr> <th rowspan="2"></th> <th colspan="5">D-R</th> </tr> <tr> <th>q_e</th> <th>β</th> <th>E_{D-R}</th> <th>R^2</th> <th>P</th> </tr> </thead> <tbody> <tr> <td>PVC-S</td> <td>20.4</td> <td>3.55E-6</td> <td>0.376</td> <td>0.949</td> <td>1.84E-5</td> </tr> <tr> <td>PVC-L</td> <td>17.1</td> <td>4.69E-6</td> <td>0.327</td> <td>0.951</td> <td>1.83E-5</td> </tr> </tbody> </table>		Linear			Langmuir				k_d	R^2	P	q_e	k_L	R^2	P	PVC-S	1.35	0.856	0.0028	42.1	0.061	0.915	6.00E-4	PVC-L	1.05	0.865	0.0024	40.2	0.044	0.909	7.52E-4		Freundlich				Temkin				k_F	n	R^2	P	B_T	k_T	R^2	P	PVC-S	3.01	1.38	0.892	1.17E-4	6.77	0.974	0.878	1.59E-4	PVC-L	2.05	1.30	0.890	1.34E-4	5.77	0.803	0.866	2.19E-4		D-R					q_e	β	E_{D-R}	R^2	P	PVC-S	20.4	3.55E-6	0.376	0.949	1.84E-5	PVC-L	17.1	4.69E-6	0.327	0.951	1.83E-5
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Reason for interest	Contained information about the interaction of microplastics with other types of pollution																																																																																									
Notes	<ul style="list-style-type: none"> - Microplastics can absorb other pollutants which is more toxic to the environment - When salinity increased, adsorption also increased - Salting out and electrostatic interaction allowed for the increase adsorption 																																																																																									
Follow up Questions	<p>To what extent does this trend apply to other pollutants and microplastics?</p> <p>Is there any combination of microplastic and pollutant that will do more damage to the environment or people than others?</p>																																																																																									

Article #8 Notes: Water Filter

Article notes should be on separate sheets

Source Title	Water Filter
Source Author	Theodore L Wilkinson and Frank J Sork
Source citation	Wilkinson, T. L., & Sork, F. J. (n.d.).
Original URL	https://patentimages.storage.googleapis.com/03/7b/45/d9954e68f49052/US5149437.pdf
Source type	patent
Keywords	
Summary of key points	<ul style="list-style-type: none"> - One end for water to enter, the other end for filtered water to come out - 3 layer of purification - Layer one: metallic particles - Layer two: activated carbon - Layer three: weak acid

Important Figures



		-continued		
Contaminant	Max. Allowed	Influent	Effluent	
Selenium	<0.01	0.09	0.007	
Mercury	<0.002	0.008	<0.0005	
Endrin	<0.0002	0.0008	<0.0002	
Lindane	<0.004	0.013	0.0014	
Methoxychlor	<0.1	0.12	0.008	
Toxaphene	<0.005	0.02	<0.005	
2,4-D	<0.1	0.19	0.017	
Silvex (2,4,5-TP)	<0.01	0.02	0.008	
TEST B (150% OF CAPACITY)				
Chloroform	<0.1	0.440	0.042	
Lead	<0.025	0.141	0.005	
Fluoride	<1.4	7.82	0.72	
Nitrate	<10.0	27.8	9.48	
Barium	<1.0	8.7	0.88	
Arsenic	<0.05	0.25	0.011	
Cadmium	<0.01	0.03	0.009	
Chromium VI	<0.05	0.14	0.024	
Chromium III	<0.05	0.162	0.013	
Selenium	<0.01	0.11	0.012	
Mercury	<0.002	0.005	<0.0005	
Endrin	<0.0002	0.0006	0.0002	
Lindane	<0.004	0.015	0.002	
Methoxychlor	<0.1	0.34	0.031	
Toxaphene	<0.005	0.018	<0.005	
2,4-D	<0.1	0.24	0.028	
Silvex (2,4,5-TP)	<0.01	0.016	0.008	
Reason for interest	Gives insight into how other filters are already filtering water			
Notes	<ul style="list-style-type: none"> - Drinking water can contain various microorganisms and other suspended particles - Activated carbon can become ineffective quickly - Anions and cations have been used in filters along with activated carbon 			
Follow up Questions	<p>What would be the average amount of usage one could get out of a water filter?</p> <p>What can cause that number to change?</p>			

Article #9 Notes: 3 - Review of Characteristics of Common Plastics for Thermoforming

Article notes should be on separate sheets

Source Title	Thermoforming of Single and Multiplayer laminates
Source Author	Seyd Ali Ashter
Source citation	Ashter, S. A. (2014). 3 - Review of Characteristics of Common Plastics for Thermoforming. In <i>Thermoforming of Single and Multilayer Laminates</i> (pp. 39–63). doi: https://doi.org/10.1016/B978-1-4557-3172-5.00003-7
Original URL	https://pdf.sciencedirectassets.com/305821/3-s2.0-C20120028219/3-s2.0-B9781455731725000037/main.pdf?X-Amz-Date=20191020T225051Z&X-Amz-Algorithm=AWS4-HMAC-SHA256&X-Amz-Signature=124b9ed3add37329dae9f64d422b12a8e5f090591bf26a53619ed53befaa1b7c&X-Amz-Credential=ASIAQ3PHCVTY45H3HCQO%2F20191020%2Fus-east-1%2Fs3%2Faws4_request&type=client&tid=pr-49591a56-fe70-4982-809f-d51eaaae6fe7&sid=dc4681ab9d1df844e88bb535fc5534831d2agxrqa&pii=B9781455731725000037&X-Amz-SignedHeaders=host&X-Amz-Security-Token=AgoJb3JpZ2luX2VjEM7%2F%2F%2F%2F%2F%2F%2F%2F%2F%2FwEaCXVzLWVhc3QtMSJGMEQCIEX1WHF7UYUm4KmU16MfrR%2FxtYUj1SZ%2FycvOx%2BWFnhY3AiBqLNbPtpf4R7Xp5c2DWj6jaS46SZYBXQWWWjCVr55zSirjAwjH%2F%2F%2F%2F%2F%2F%2F%2F%2F%2F8BEAlaDDA1OTAwMzU0Njg2NSIM%2FjOLPxy3GFGlJl3VKrcD2NXaMjNjF0JzBKOvsvHcHr215aK5KtkQPIvt644rQgnD9ZwtsNjvEA%2BmZ5ViCEAHaYKo%2F5s3Bu%2BjBzMNuUX7ARdZ8%2FjCUzR0j7E7f4cgJwZWZO9kJpO1BFADiv7X10iAwzFkaxFiO3SVCSCTUasVOtSVYKvMSJ5mRxk30DyQUlczfKNKz6DYE%2Bt6z22uyhItaRyw6z1tmZ6TrSrvlPpd4uZUhx5YmucoNNalr9pQHS6jlWtbwKFWjeA1hJctq31qmHQJI000McQV0pu7nSSVHiKxQEwaME%2B6f4NTJCer2iJxccUx7bo9SYf4KqF0H%2Fog9dhjm0rYoFy0nrb8WwsOu4VDR%2FzbT2YGimbkaMQm1LWqCZlbp0R5zw5Sj3hhUc8medRzthi02GDQlpXhJmqiCupdiKh5iJYqsGyzzlvQPLQ8edGC21ygrzpz8t0GLEdlnZ2PUeFQSjTCYwUY1%2B%2FTK9XGhiX%2F6IP6BX9YswpWp5DWk7izCkAJ08%2Bd96MMjSMmuO9qN6Cqf6djr3lBuQxg98zF1q439Q%2BtCp%2FeAdhSIS2BNpM3kgYuZnJfUAm%2BrHnx9mu42w4l%2BzCzpbPtBTq1AZLtTBx3NnYN%2B5lvEpod2wCLELYCY6Y9XJCih0LYqf4sjcGeMjiBxPhi6BtQMJoZb7LzL3rxfvXunIO0Bctc5Gm7m3G44fOuvX3aZlbp7BVgBKxovl4LgpTkoCfL6JKfPqa2biPB3ltMRtma5tTGJnRV6K4jSFVtAC0Xnzs04bFZZeLwvyD%2FEJfw2krHsdLsF1kiUej%2F7aw1xFj76qo

	%2BNqskvbdaUDBKXAMWxyGDESy1la%2FEo%3D&host=68042c943591013ac2b2430a89b270f6af2c76d8dfd086a07176afe7c76c2c61&X-Amz-Expires=300&hash=c1d73a84f0a7fe25dfe905e0c98cd8c659959acf350c0943012a4e7af3f180be																																																			
Source type	Book Chapter																																																			
Keywords	Crystallinity, glass transition, temperature viscosity, molecular weight, molecular weight, distribution, electrical resistance, chemical resistance, mechanical properties																																																			
Summary of key points	<ul style="list-style-type: none"> - Some plastics are tough and don't break easily - Some are very tough but are brittle - Different plastics will have different properties and will come with advantages and disadvantages 																																																			
Important Figures	<p>Table 3.2 Comparison of Mechanical Properties of Polyamide Family [16]</p> <table border="1"> <thead> <tr> <th>Polyamide Type</th> <th>Specific gravity (g/cm³)</th> <th>Tensile Strength (MPa)</th> <th>Tensile Modulus (MPa)</th> <th>Flexural Modulus (MPa)</th> <th>Elongation at break (%)</th> <th>Izod impact strength (kJ/m²)</th> </tr> </thead> <tbody> <tr> <td>Nylon 6</td> <td>1.13</td> <td>70</td> <td>2.8</td> <td>2.2</td> <td>15</td> <td>45–65</td> </tr> <tr> <td>Nylon 6, 6</td> <td>1.14</td> <td>85</td> <td>3.0</td> <td>2.8</td> <td>5</td> <td>40–60</td> </tr> <tr> <td>Nylon 6, 10</td> <td>1.07</td> <td>55</td> <td>2.1</td> <td>–</td> <td>70</td> <td>50</td> </tr> <tr> <td>Nylon 11</td> <td>1.03</td> <td>38</td> <td>1.4</td> <td>1.2</td> <td>250</td> <td>100</td> </tr> </tbody> </table> <p>Table 3.3 Mechanical Properties of PTFE [27]</p> <table border="1"> <thead> <tr> <th>Property</th> <th>Value</th> </tr> </thead> <tbody> <tr> <td>Tensile strength, MPa</td> <td>20–35</td> </tr> <tr> <td>Elongation at break, %</td> <td>300–550</td> </tr> <tr> <td>Tensile modulus, MPa</td> <td>550</td> </tr> <tr> <td>Flexural strength, MPa</td> <td>No break</td> </tr> <tr> <td>Flexural modulus at 23 C, MPa</td> <td>340–620</td> </tr> <tr> <td>Impact strength, Izod, notched, J/m</td> <td>188</td> </tr> <tr> <td>Compressive strength, MPa</td> <td>34.5</td> </tr> </tbody> </table>	Polyamide Type	Specific gravity (g/cm ³)	Tensile Strength (MPa)	Tensile Modulus (MPa)	Flexural Modulus (MPa)	Elongation at break (%)	Izod impact strength (kJ/m ²)	Nylon 6	1.13	70	2.8	2.2	15	45–65	Nylon 6, 6	1.14	85	3.0	2.8	5	40–60	Nylon 6, 10	1.07	55	2.1	–	70	50	Nylon 11	1.03	38	1.4	1.2	250	100	Property	Value	Tensile strength, MPa	20–35	Elongation at break, %	300–550	Tensile modulus, MPa	550	Flexural strength, MPa	No break	Flexural modulus at 23 C, MPa	340–620	Impact strength, Izod, notched, J/m	188	Compressive strength, MPa	34.5
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Reason for interest	It talks about the properties of the most common plastics which may help in the filtering process.																																																			
Notes	<ul style="list-style-type: none"> - Polyester is used in the production of film and fibers - Polyethylene terephthalate is hard, strong and extremely tough, has high abrasion resistant, and chemical resistant with low moisture absorption. It is used in special areas like electronics components, film and drinking bottles. - Polyethylene has a density less than that of water. - It is tough but has moderate tensile strength 																																																			

	- The molecular weight can range from 10,000 to 40,000
Follow up Questions	Which plastics would pose the most issue with filtering?

Article #10 Notes: A simple method for detecting and quantifying microplastics utilizing fluorescent dyes - Safranin T, fluorescein isophosphate, Nile red based on thermal expansion and contraction property

Article notes should be on separate sheets

Source Title	Environmental pollution
Source Author	Lulu Lv, Junhao Qu, Zihua Yu, Daihuan Chen, Pengzhi Hong, Shengli Sun, Chenyong Li
Source citation	Lv, L., Qu, J., Yu, Z., Chen, D., Hong, P., Sun, S., & Li, C. (n.d.). A simple method for detecting and quantifying microplastics utilizing fluorescent dyes - Safranin T, fluorescein isophosphate, Nile red based on thermal expansion and contraction property. <i>Environmental Pollution</i> , 225. doi: https://doi.org/10.1016/j.envpol.2019.113283
Original URL	https://www.sciencedirect.com/science/article/pii/S0269749119329562
Source type	Journal
Keywords	Microplastics, Staining, Thermal expansion and contraction, Identification, Quantification
Summary of key points	<ul style="list-style-type: none"> - The method of staining was good for detecting polyethylene, polystyrene, polyvinyl chloride, and polyethylene terephthalate.

Important Figures		Table 2 The recovery data of four polymers stained with Nile red, FITC and Safranin T under different heating conditions (The numerator indicates the number of microplastics fluorescent particles under fluorescence field, and the denominator represents the number of microplastics particles in the bright field).									
		Microplastics									
		Dyes									
		Nile red Time (min)			FITC Time (min)			Safranin T Time (min)			
		10	20	30	10	20	30	10	20	30	
PE Temperature (°C)	25	18/20	21/22	21/22	18/32	22/26	27/34	0/24	0/25	0/28	
	50	30/30	25/25	24/24	22/22	24/24	31/31	28/28	29/29	23/23	
	75	25/25	23/23	26/26	20/21	20/21	22/24	20/20	24/24	28/28	
PVC Temperature (°C)	25	35/35	31/31	34/34	29/35	26/26	25/25	0/28	0/30	0/28	
	50	28/28	32/32	29/29	27/29	26/30	32/34	26/27	31/31	28/28	
	75	21/21	25/25	37/37	27/28	26/28	25/25	32/32	25/25	33/33	
PET Temperature (°C)	25	19/24	29/31	26/32	20/26	22/26	20/20	0/30	0/28	0/21	
	50	24/24	24/24	30/30	20/27	23/26	19/26	28/39	23/28	18/21	
	75	21/21	24/24	32/32	20/23	20/23	21/24	25/29	25/31	24/33	
PS Temperature (°C)	25	8/31	8/30	11/27	19/22	26/30	20/23	0/21	0/23	0/20	
	50	30/30	29/29	26/26	26/26	29/29	27/27	21/23	30/31	21/23	
	75	23/23	30/30	32/32	25/27	21/23	29/31	20/22	22/26	23/23	

Table 3 Recovery rates of four microplastics stained with three dyes (in triplicate and based on weight calculations). Staining conditions are 50 °C for 30 min.				
Microplastics	Dyes			
	Nile red Recovery rates (%)	FITC Recovery rates (%)	Safranin T Recovery rates (%)	
PE	96 ± 2	98 ± 0.2	96 ± 0.2	
PVC	93 ± 1	95 ± 0.5	98 ± 3	
PET	96 ± 0.5	97 ± 0.5	99 ± 0.5	
PS	96 ± 0.4	98 ± 0.3	99 ± 0.2	

Reason for interest	There may be some properties more specific to microplastics due to their size.
Notes	<ul style="list-style-type: none"> - Visually identifying microplastics is not a good way to observe them. - Certain dyes can only be absorbed by certain microplastics. - Red Nile could only stain polyethylene, polystyrene, polypropylene, and nylon 6.
Follow up Questions	Can I use this method in my experiments? Is it costly to procure these materials.

Article #11 Notes: Dangerous hitchhikers? Evidence for potentially pathogenic *Vibrio* spp. on microplastic particles

Article notes should be on separate sheets

Source Title	Dangerous hitchhikers? Evidence for potentially pathogenic <i>Vibrio</i> spp. on microplastic particles
Source Author	Igna V. Kirstein, Sidika Kirmizi, Antje Wichels, Alexa Garin-Fernandez, Rene Erler, Martin Loder, Gunnar Gerdt
Source citation	Kirstein, I. V., Kirmizi, S., Wichels, A., Garin-Fernandez, A., Erler, R., Loder, M., & Gerdt, G. (2016). Dangerous hitchhikers? Evidence for potentially pathogenic <i>Vibrio</i> spp. on microplastic particles. <i>Marine Environmental Research</i> , 120, 1–8. doi: https://doi.org/10.1016/j.marenvres.2016.07.004
Original URL	https://www.sciencedirect.com/science/article/pii/S014111361630112X
Source type	Journal
Keywords	Synthetic polymers, Vector, Pathogens, North Sea, Baltic Sea
Summary of key points	
Important Figures	<p>i. <small>I.V. Kirstein et al. / Marine Environmental Research 120 (2016) 1–8</small></p> <p>The figure consists of three donut charts representing the composition of microplastic particles. The first chart (left) is for RV Heincke in September 2013 (N=103), showing 55% Polyethylene, 17% Polypropylene, and 7% Polystyrene. The second chart (middle) is for Helgoland in August 2013 (N=15), showing 40% Polyethylene, 20% Polypropylene, and 7% Polystyrene. The third chart (right) is for RV Heincke in July/August 2014 (N=82), showing 40% Polyethylene, 14% Polypropylene, and 5% Polystyrene. A legend below the charts identifies the materials: Polyethylene (dark blue), Polypropylene (red), Polystyrene (green), Varnish (yellow), Acrylnitril-Butadien-Styrole (light blue), Polyvinyl alcohol (orange), Polyamide (black), Ethylene vinyl alcohol (light red), Polyurethane (light green), Polyvinylchloride (purple), Keratin (grey), and Stearic Acid (light blue-grey). 'Not Identified' is also listed in the legend.</p>

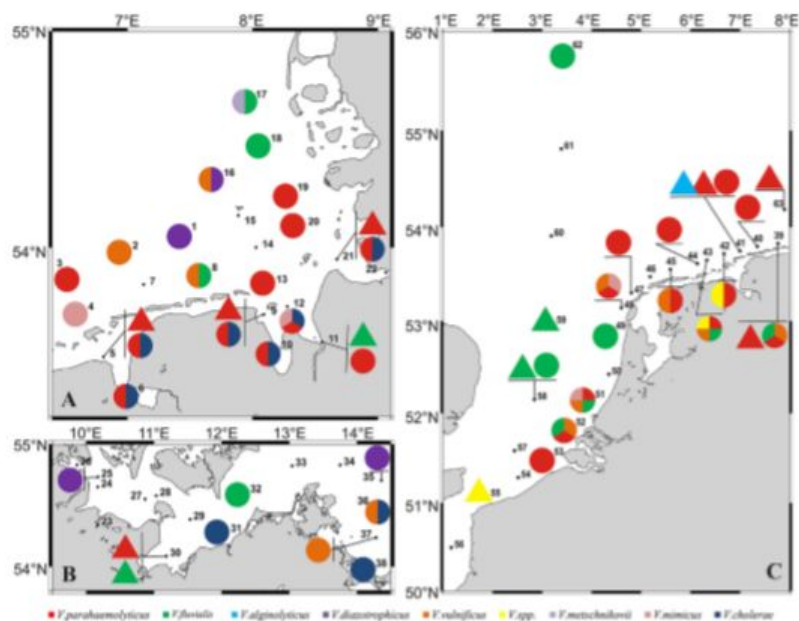


Fig. 2. Geographical occurrence of *Vibrio* spp. On microplastics and surface water of a) the North Sea from research cruise HE409 on RV Heincke in September 2013 b) the Baltic Sea from research cruise HE409 on RV Heincke in September 2013 and c) North Sea from research cruise HE430 on RV Heincke in July/August 2014 and the drift line of Helgoland station 63. (○) species detected from surrounding seawater (△) species detected on microplastic particles.

Reason for interest	This article lists some of the dangers of the microplastics in the environment.
Notes	
Follow up Questions	