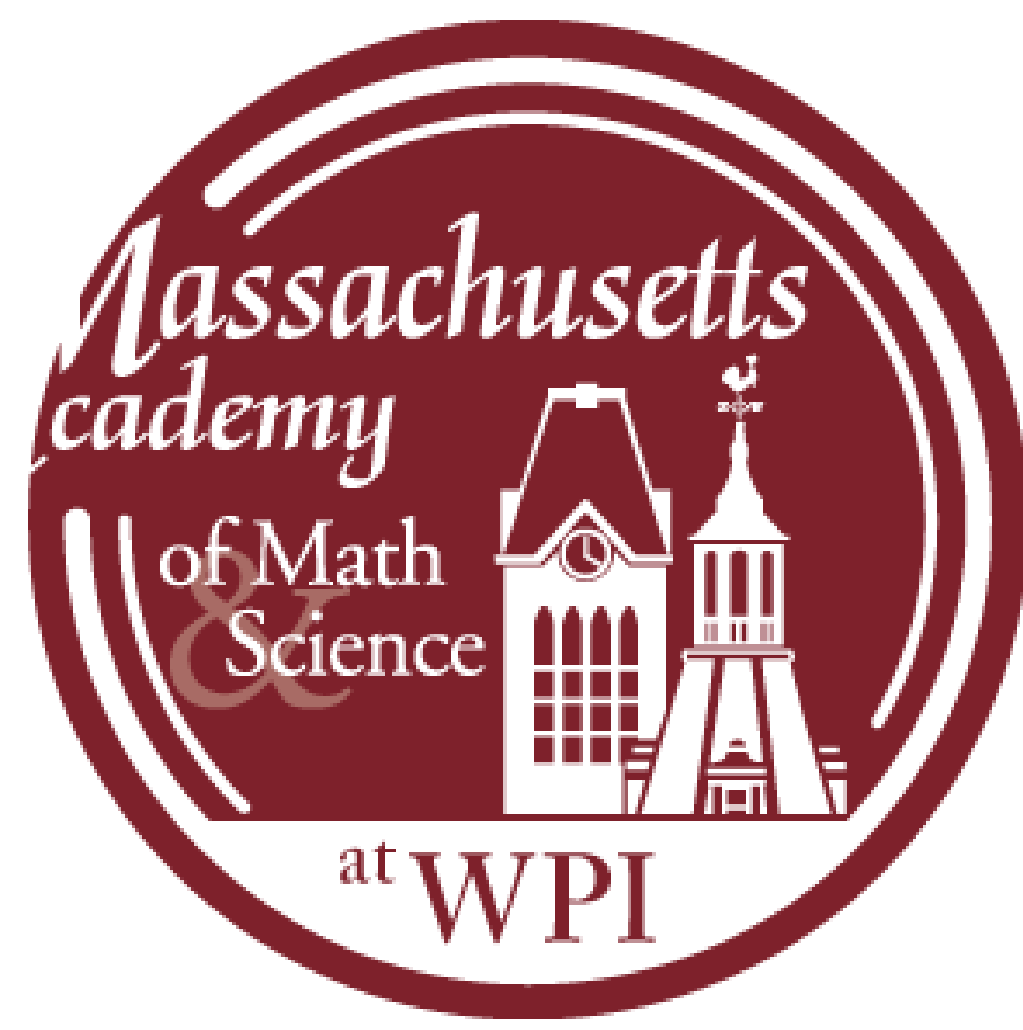




WPI



CO2 Filtration Device for Automobile Exhausts

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11 SUSTAINABLE CITIES AND COMMUNITIES



12 RESPONSIBLE CONSUMPTION AND PRODUCTION



13 CLIMATE ACTION



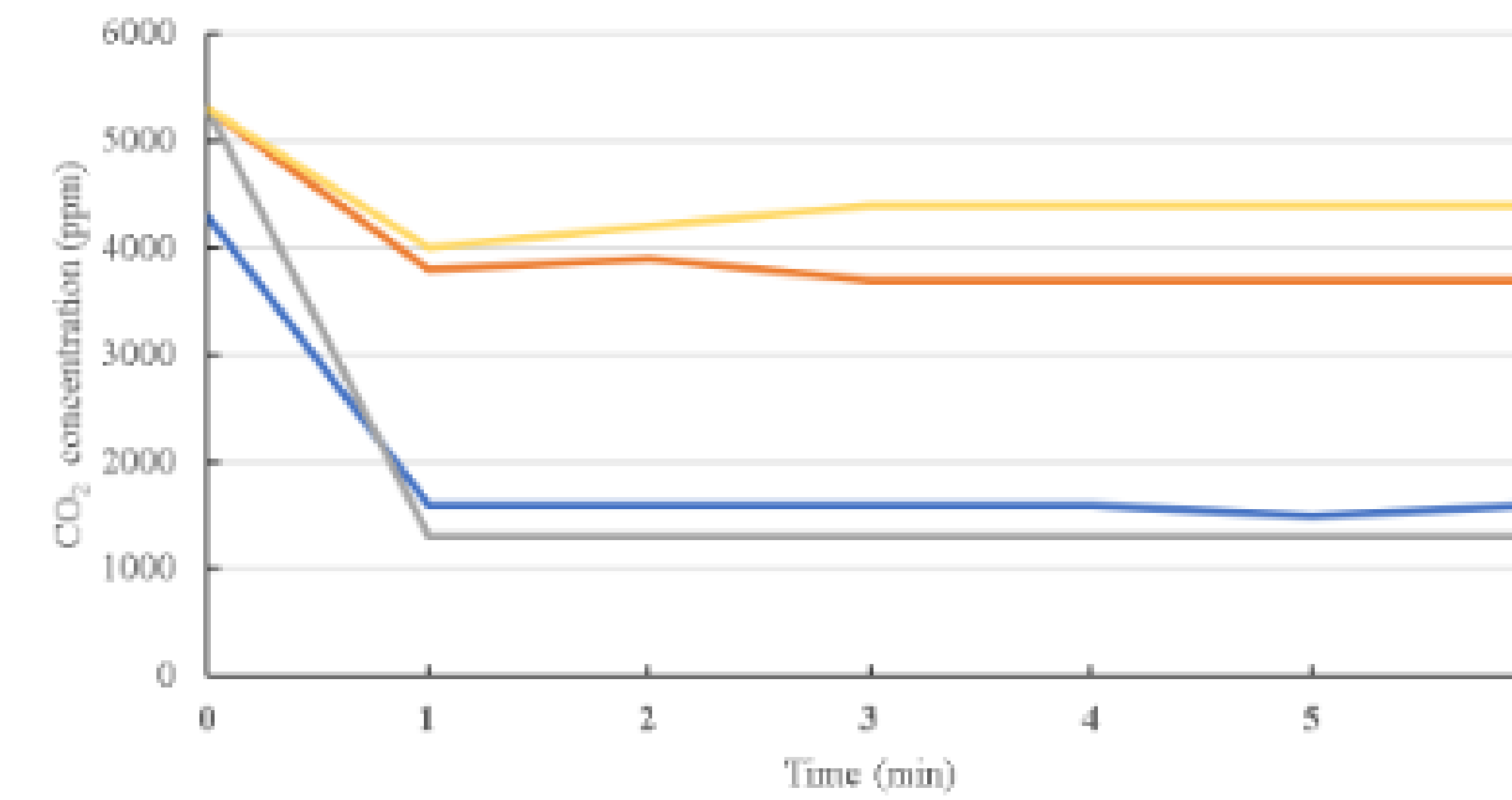
Engineering Need

There is currently little **consumer-level technology** that directly captures or removes CO₂ at the **point of emission** in cars.

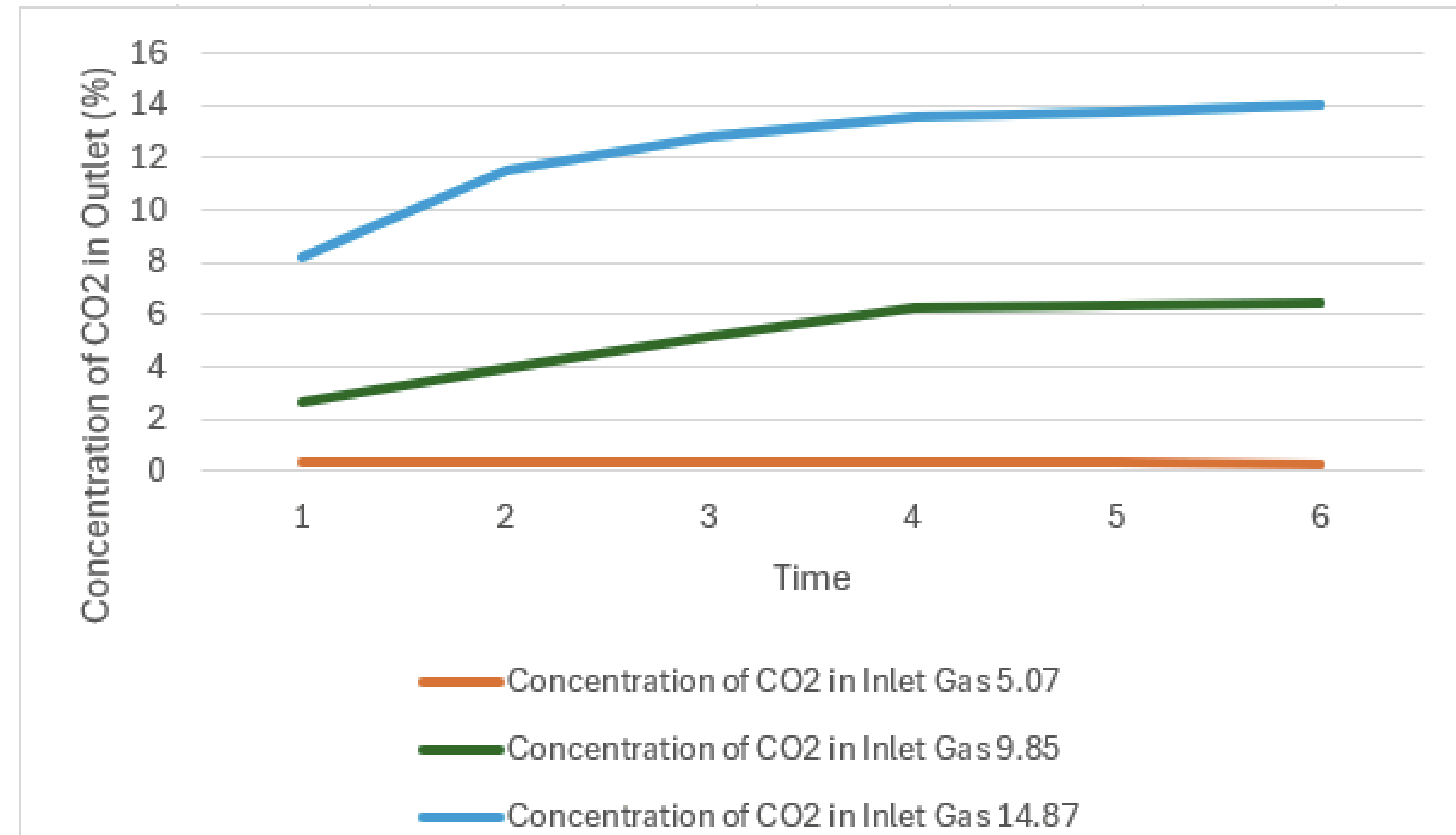
Engineering Objective

Design and prototype a chemical-scrubbing exhaust attachment that uses **Hydroxide-based filtration** to remove CO₂ from **car exhaust streams**

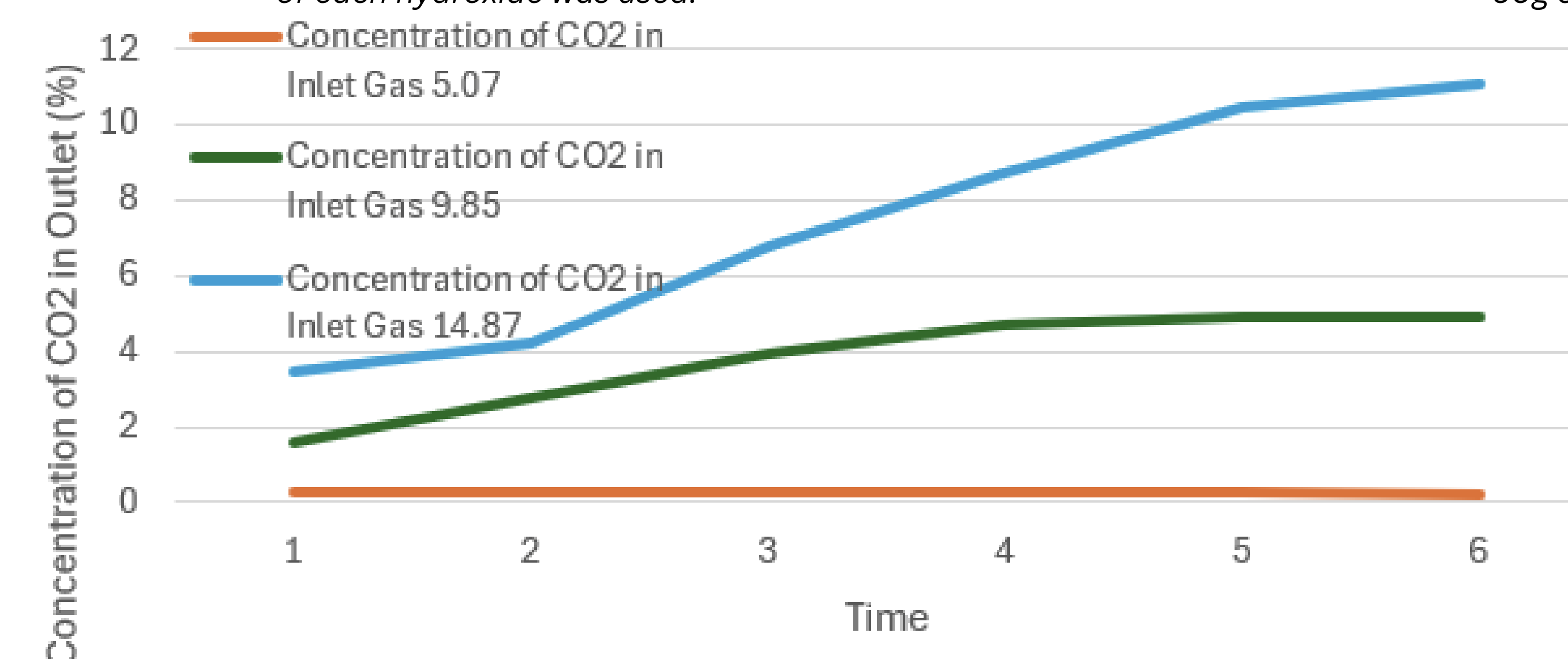
Results/Figures



Graph 1: Hydroxide Tests. Shows concentration of CO₂ (ppm) in outlet gas after passing through the different hydroxides, over 6 minutes. 25g of each hydroxide was used.



Graph 2: Concentration test. Shows concentration of CO₂ in outlet gas after passing through NaOH, with varying initial CO₂ concentrations. 50g of hydroxide was used.



Graph 3: Concentration test with Dual impingers. Shows concentration of CO₂ in outlet gas after passing through 2 layers of NaOH, with varying initial CO₂ concentrations. Uses 50g per impinger.

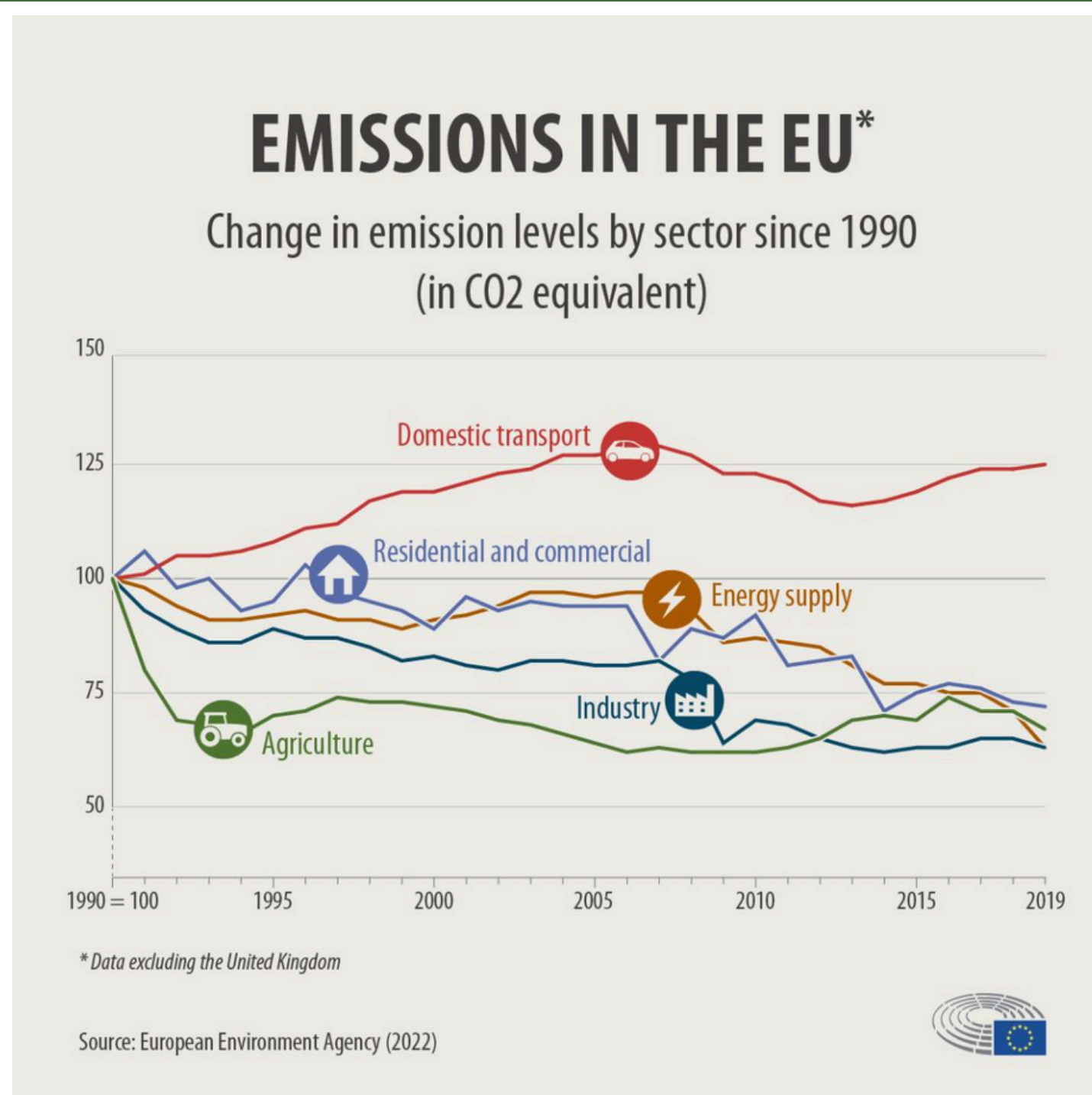


Figure 1: Compares changes in emission levels of different sectors in the EU since 1990. (Parliament, 2019)

- Driving **Climate Change**
- **Transportation** is a large portion of CO₂ emissions
- **33.5%** increase in emissions between 1990 and 2019 (Parliament, 2019)
- **Hydroxides** - Chemicals containing hydroxide ion (OH)
- **Chemical Scrubbing** - Filters by creating new byproducts within a chemical reaction
- **Impinger** - Device that allows airflow, and contains a filtration method

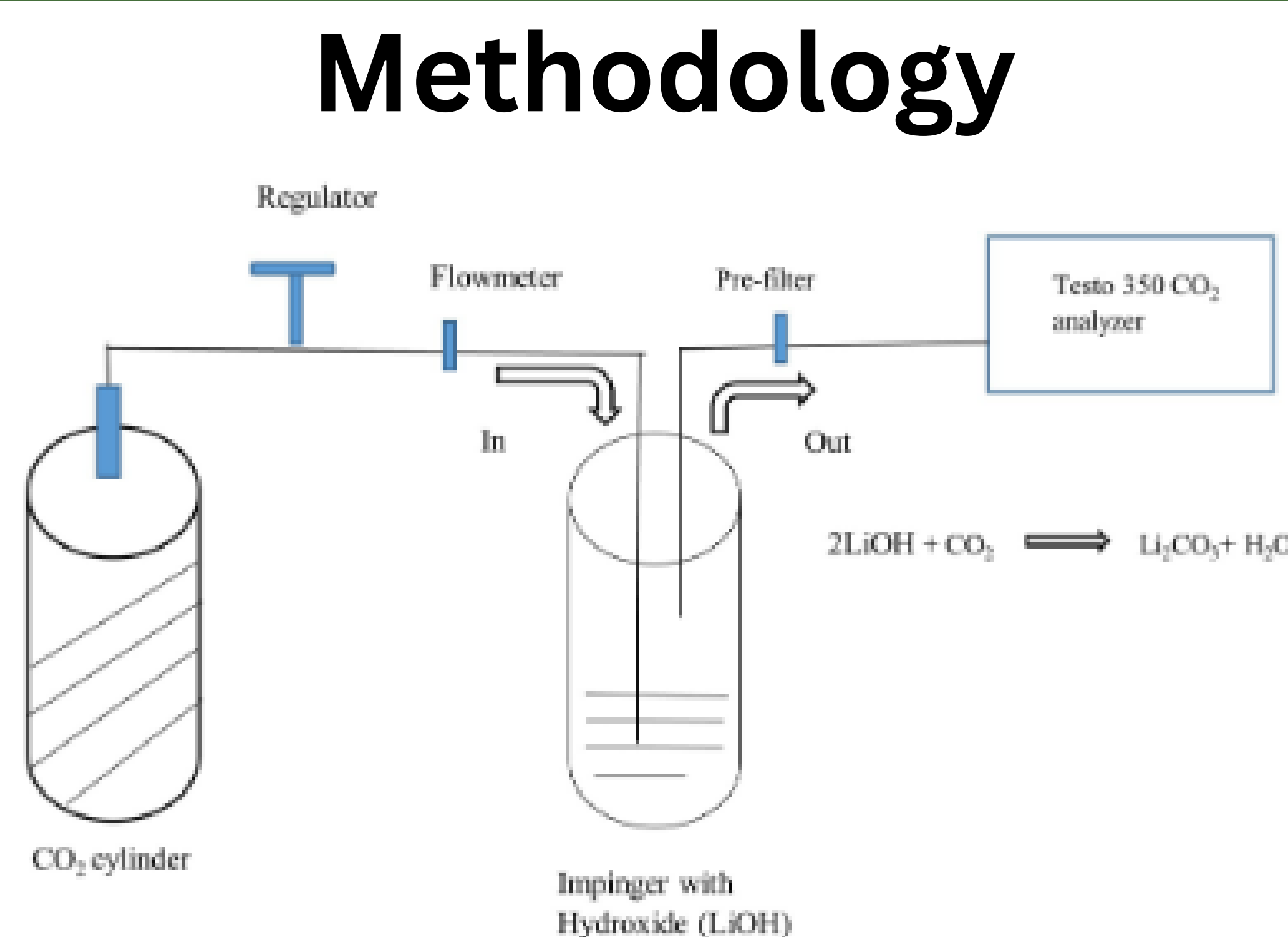


Figure 2: Hydroxide testing procedure. Using CO₂ gas, impingers and a gas analyzer. Chemical scrubbing produces Carbonate & Water. Example showcases process with Lithium Hydroxide(LiOH)

- **Hydroxide testing** - Cylinder filled with Nitrogen-gas mixture flows into impinger containing hydroxide. Filtered gas flows into gas analyzer (Graph 1)
- Using the same method, I then varied the **CO₂ concentration** of the gas and tested filtration of **1 hydroxide** (Graph 2)
- Finally I varied CO₂ concentration again, but with **two impingers** (Table 3)

- Test 1:**
- LiOH ~ 32.40%
 - KOH ~ 68.60%
 - NaOH ~ 74.5%
 - Ca(OH)₂ ~ 24.9%
- NaOH was **most efficient** at filtering CO₂ at a low level

- Analysis**
- Test 2:**
- @5% ~ 93.2%
 - @10% ~ 63.6%
 - @15% ~ 25.1%
- NaOH is **inefficient** at high concentrations

- Test 3:**
- @5% ~ 95.6%
 - @10% ~ 78.0%
 - @15% ~ 51.2%
- **Multi layered** filtration greatly increases efficiency

Future Steps

- **Improve filtration** capability with more layers/more hydroxide
- **Construct and test** physical filter device in **real-world** applications

Key Takeaways

- **NaOH** filtered CO₂ most efficiently
- **Multi-layered** hydroxide filtration improved filtration
- **Real world** building & testing is needed