

# Project Notes:

**Project Title: CO2 Filtration Device for Automobile Exhausts**

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**Note Well:** There are NO SHORT-cuts to reading journal articles and taking notes from them. Comprehension is paramount. You will most likely need to read it several times, so set aside enough time in your schedule.

## **Contents:**

Knowledge Gaps:	2
Literature Search Parameters:	3
Tags:	3
Article #0 Notes: Title	4
Article #1 Notes: Low carbon optimization for wind integrated power systems with carbon capture and energy storage under carbon pricing *Formal Article*	5
Article #2 Notes: Progress in carbon capture technologies	14
Progress in carbon capture technologies	14
Article #3 Notes: Carbon capture technology to produce clean fuel from air	16
Article #4 Notes: Carbon Capture Marches Toward Practical Use	18
Article #5 Notes: Polymeric membranes in carbon capture, utilization, and storage: current trends and future directions in decarbonization of industrial flue gas and climate change mitigation	20
Article #6 Notes: A Review of Materials for Carbon Dioxide Capture	22
Article #7 Notes: Recent advancements in carbon capture materials research: Innovative optimization of materials synthesis and engineering applications	24
Article #8 Notes: A critical appraisal of advances in integrated CO <sub>2</sub> capture and electrochemical conversion	25
Article #9 Notes: Carbon Capture by Biological Methods	27
Article #10 Notes: Carbon Capture, Utilization, and Storage in the MENA Region: A Regional Review of Projects and Challenges	29
Article #11 Notes: CO <sub>2</sub> Capture: A Comprehensive Review and Bibliometric Analysis of Scalable Materials and Sustainable Solutions	31
Article #12 Notes: An ecosystem of carbon dioxide removal reviews – part 1: direct air CO <sub>2</sub> capture and storage	33
Article #13 Notes: Electrochemical CO <sub>2</sub> capture with pH-independent redox chemistry	34

Article #14 Notes: The Open DAC 2025 Dataset for Sorbent Discovery in Direct Air Capture	36
Article #15 Notes: Design and demonstration of a direct air capture system with moisture-driven CO <sub>2</sub> delivery into aqueous medium	37
Article #16 Notes: Enhancing Direct Air Capture through Potassium Carbonate Doping of Activated Carbons	39
Article #17 Notes: Intrinsic Direct Air Capture	40
Article #18 Notes: Carbon capture and recycling technology of carbon resources under the target of carbon neutrality	41
Article #19 Notes: A comprehensive review of enhanced CO <sub>2</sub> capture using activated carbon derived from biomass feedstock	43
Article #20 Notes: Revolutionizing carbon capture efficiency: A comprehensive review of AI-Driven approaches	44



## Knowledge Gaps:

This list provides a brief overview of the major knowledge gaps for this project, how they were resolved and where to find the information.

Knowledge Gap	Resolved By	Information is located	Date resolved
Current Carbon Capture Technologies	Reading article about Current Progress on Carbon Capture	<a href="https://www.sciencedirect.com/science/article/pii/S0048969720367346">https://www.sciencedirect.com/science/article/pii/S0048969720367346</a>	9/28/2025
Current most effective CC	Reading a review paper global Carbon capture use	<a href="https://www.proquest.com/pq1sustainability/docview/3180988994/8761899E96BE48BBPQ/7?accountid=29120&amp;sourcecetype=Conference%20Papers%20&amp;%20Proceedings">https://www.proquest.com/pq1sustainability/docview/3180988994/8761899E96BE48BBPQ/7?accountid=29120&amp;sourcecetype=Conference%20Papers%20&amp;%20Proceedings</a>	9/28/2025
Current cheapest CC	Reading a review paper global Carbon capture use	<a href="https://www.proquest.com/pq1sustainability/docview/3180988994/8761899E96BE48BBPQ/7?accountid=29120&amp;sourcecetype=Conference%20Papers%20&amp;%20Proceedings">https://www.proquest.com/pq1sustainability/docview/3180988994/8761899E96BE48BBPQ/7?accountid=29120&amp;sourcecetype=Conference%20Papers%20&amp;%20Proceedings</a>	9/28/2025



## Literature Search Parameters:

These searches were performed between 9/1/2025 and XX/XX/2026.

List of keywords and databases used during this project.

Database/search engine	Keywords	Summary of search
Environmental Science Database	Carbon Capture	Searched for articles and papers relating to Carbon Capture technology.
Nature	Carbon Capture + Renewable Energy	Articles relating to CC in combination with RE, specifically Article #1.
Environmental Science Database + Nature + Engineering Village	Carbon Capture + Drone	I couldn't find any experiments done about combining carbon capture technology into a mobile drone.

## Tags:

Tag Name	



## Article #0 Notes: Title

Article notes should be on separate sheets

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<b>Source Title</b>	
<b>Source citation (APA Format)</b>	
<b>Original URL</b>	
<b>Source type</b>	
<b>Keywords</b>	
<b>#Tags</b>	
<b>Summary of key points + notes (include methodology)</b>	
<b>Research Question/Problem/ Need</b>	
<b>Important Figures</b>	
<b>VOCAB: (w/definition)</b>	
<b>Cited references to follow up on</b>	
<b>Follow up Questions</b>	

# Article #1 Notes: Low carbon optimization for wind integrated power systems with carbon capture and energy storage under carbon pricing \*Formal Article\*

Article notes should be on separate sheets

<b>Source Title</b>	Low carbon optimization for wind integrated power systems with carbon capture and energy storage under carbon pricing
<b>Source citation (APA Format)</b>	Meng, Q., He, Y., Hussain, S., Lu, J., & Guerrero, J. M. (2025). Low carbon optimization for wind integrated power systems with carbon capture and energy storage under carbon pricing. <i>Scientific Reports</i> , 15(1), 32714. <a href="https://doi.org/10.1038/s41598-025-17463-7">https://doi.org/10.1038/s41598-025-17463-7</a>
<b>Original URL</b>	<a href="https://doi.org/10.1038/s41598-025-17463-7">https://doi.org/10.1038/s41598-025-17463-7</a>
<b>Source type</b>	Scientific Article
<b>Keywords</b>	Wind Power; Carbon Capture; Pre-Combustion; Post-Combustion; Energy Storage System; Cost efficiency
<b>#Tags</b>	#System #Model
<b>Summary of key points + notes (include methodology)</b>	<ul style="list-style-type: none"> <li>• Wind power enables low-carbon energy transition</li> <li>• Model evaluates impact of CC prices on energy storage &amp; unit power supply costs under high winds</li> <li>• Increased wind power capacity = cheaper cost of electricity, and improves stability with energy storage</li>   <li>• Wind power plays a critical role in “decarbonizing electricity generation”</li> <li>• Current wind power utilization is unreliable and unstable</li> <li>• Because wind power isn’t constantly available, energy is wasted, and we instead rely on fossil fuels. <ul style="list-style-type: none"> <li>○ Energy Storage Systems mitigate these negative effects</li> <li>○ However, ESS are costly and not easy to use at large volumes</li> </ul> </li> <li>• Combine CCS and ESS to control carbon emissions, and optimize storage utilization &amp; renewable penetration <ul style="list-style-type: none"> <li>○ However, current models don’t do this.</li> </ul> </li> <li>• Optimization of wind energy systems is currently done by analyzing wind patterns and attempting to adjust, but these are still limited: <ul style="list-style-type: none"> <li>○ Isolate economic or stability goals</li> <li>○ Limited consideration of carbon pricing impacting mechanisms.</li> <li>○ GAs are also insufficient</li> </ul> </li> <li>• Critical gap = lack of models that unify CCS, ESS and wind power</li> </ul>

- Current frameworks don't quantify how CC prices influence storage or power generation costs.
- The study proposes a model that includes wind power, CCS, and ESS within a unified "optimization framework".
  - CCS & ESS work together to optimize CC charging cycles, saving 22% more power.
  - Links carbon prices to optimal ESS capacity revealing nonlinear cost-emission trade-offs.
  - Improved GA (an algorithm that solves for the best possible solution) has a 98% convergence (point at best solution) reliability
- Increasing wind power usage from 30 to 50% reduces electricity costs by 19% while ESS is deployed at 127MWh at carbon price of 100 CNY per ton, producing 75 MW power output
  - This is one optimal design the researchers came up with using their model, that when carbon price is 100CNY, the most cost-effective low-carbon way to tune the model is to build a system that stores 127 MWh (ESS), which will then deliver 75 MW of power, and reduce costs by 19%.
- Quantifying carbon price creates actionable insights
- Improved GA allows for function on a larger scale.
- CCS + ESS combination creates a good framework for targeting net-zero emissions.

## Section 2

- Low-carbon optimization model for power systems with high wind power usage.
  - Incorporates CC to reduce carbon emissions and enhance WP generation
  - Analyzes carbon capture costs and all other costs (operating, investment, etc.) to find optimal allocation of energy storage.
  - Uses arithmetic examples and an improved GA to solve
- Through all of this, the model derives the optimal capacity allocation.
- Post-combustion decarbonization proved to be the most efficient at a relatively low cost, so that is what is used in the model.
- Used to find power used by CC device. ( $P$ =input power,  $P_{CO_2}$ =energy consumed by CC device):

$$P = P_{CO_2} + P_e$$

$$P_{CO_2} = P_{run} + P_s$$

○

- Used to find power output of wind turbines. ( $P_w$ =output power of fan,  $P_r$ =rated power of fan)

$$P_w(v) = \begin{cases} 0, v < v_{in} \text{ or } v > v_{out} \\ P_r \frac{v - v_{in}}{v_r - v_{in}}, v_{in} \leq v \leq v_r \\ P_r, v_r \leq v \leq v_{out} \end{cases}$$

$$P_r = \frac{1}{2} \pi r^2 \rho v^3 C_p$$

- 
- Power used by Thermal Power units ( $F_i$  = Coal consumption,  $P_{Gi,down}$  = Lower limit of crepage rate of thermal unit,  $U_{Gi}$  = Status marker of Thermal unit)

$$F_i(t) = a_i P_{Gi}^2(t) + b_i P_{Gi}(t) + c_i$$

○

$$U_{Gi}(t) P_{Gi, \min} \leq P_{Gi}(t) \leq U_{Gi}(t) P_{Gi, \max}$$

$$P_{Gi, \downarrow} \leq P_{Gi}(t) - P_{Gi}(t-1) \leq P_{Gi, \uparrow}$$

○

- Charge and discharge of battery

$$\begin{aligned} U_{bt, ch}(t) P_{bt, ch, \min} &\leq P_{bt, ch}(t) \leq U_{bt, ch}(t) P_{bt, ch, \max} \\ U_{bt, dis}(t) P_{bt, dis, \min} &\leq P_{bt, dis}(t) \leq U_{bt, dis}(t) P_{bt, dis, \max} \end{aligned}$$

○

- Load state constraints (must meet these electrical demands)

$$\begin{aligned} SOC_{bt}(t) &= SOC_{bt}(t-1) + (\eta_{bt, ch} P_{bt, ch}(t) - P_{bt, dis}(t) / \eta_{bt, dis}) \Delta t \\ SOC_{bt}^{\min} &\leq SOC_{bt}(t) \leq SOC_{bt}^{\max} \end{aligned}$$

○

- Battery degradation and life-cycle considerations are excluded. Integrating these to assess long-term economics is good ground for future work.

### Section 3

- Function representing model's goal of minimizing cost of supplying

electricity

- C = System unit cost of Supplying electricity
- $C_f$  = System unit cost of electricity supply
- $P_L$  = Total System Load

$$\min C = \frac{C_F}{P_L}$$

○

$$C_F = C_h + C_{cb} + C_{wind} + C_b + C_{qw} + C_{loss}$$

$$C_h = \sum_{t=1}^{24} \sum_{i=1}^N k_f U_{Gi}(t) F_i(t)$$

$$C_{cb} = \sum_{t=1}^{24} \sum_{i=1}^N dv_i F_i(t)$$

$$C_{wind} = \sum_{t=1}^{24} k_w P_w(t)$$

▪

$$C_{bt} = \sum_{t=1}^{24} (k_e P_{bt}(t) + k_p E_{bt}(t))$$

$$C_{qw} = \sum_{t=1}^{24} k_{qw} (P_w^{pre}(t) - P_w(t))$$

$$C_{loss\_load} = \sum_{t=1}^{24} k_{ql} \left( \sum_{i=1}^N P_{Gi}(t) + P_w(t) - P_l(t) \right)$$

▪

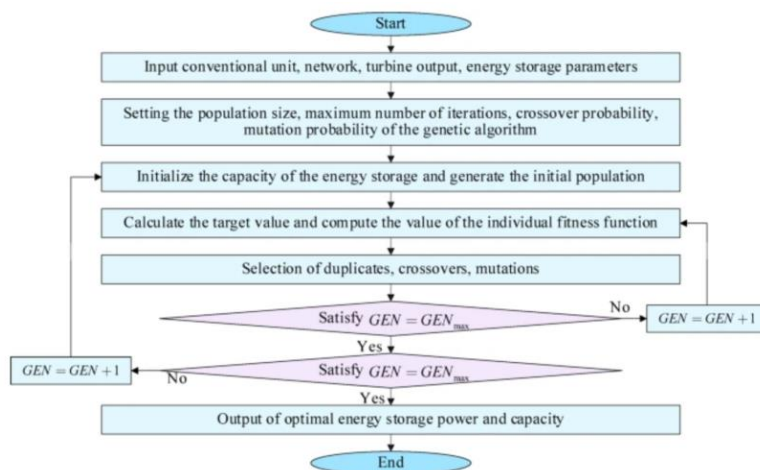
#### Section 4

- The GA is an algorithm that emulates natural selection, in which the best traits are selected, crossed over, and mutated until the algorithm's completion (when it determines the optimal outcome)
- Improved GA incorporates adaptive functions to address changing probability of crossover ( $P_c$ ) and probability of mutation ( $P_m$ )

- This makes the algorithm automatically adjust how much it explores (tries new solutions) and how much it exploits (refines old solutions) based on findings of each step.
- Explained by the expressions:

$$p_c = \begin{cases} k_1 \frac{f_{\max} - f'}{f_{\max} - f_{avg}}, & f' \geq f_{avg} \\ k_2, & f' < f_{avg} \end{cases}$$

$$p_m = \begin{cases} k_3 \frac{f_{\max} - f'}{f_{\max} - f_{avg}}, & f' \geq f_{avg} \\ k_4, & f' < f_{avg} \end{cases}$$



- The flow chart of an improved GA algorithm.

### Section 5 (Testing)

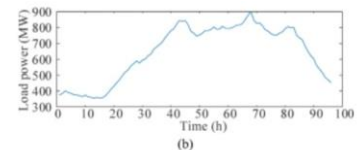
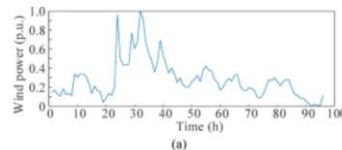
- To test, a system configuration is established:
  - 3 thermal power units
  - One wind power unit
  - One carbon capture device
  - Coal/Carbon price is 700 CNY per ton
  - Cost per unit of wind power electricity is 0.045 CNY per kWh
  - Wind abandonment = 0.3 CNY/kWh
  - Load abandonment = 8 CNY/kWh
- By using the given configurations and data, they evaluated parameters of the model:
  - Parameters relating to a Thermal Power Unit

Technical Parameters	Unit 1	Unit 2	Unit 3
Maximum technical output of the unit (MW)	600	300	150
Minimum technical output of the unit (MW)	180	90	45
Carbon emissions (kg/kWh)	0.72	0.75	0.79
Coal consumption factor c (kg/h)	786.8	451.32	1049.5
Coal consumption factor b (kg/MWh)	30.42	65.12	139.6
Coal consumption factor a (kg/MWh)	0.226	0.588	0.785

- Parameters relating to Energy Storage

Norm	Parameters
Unit power cost (CNY/kW)	3000
Unit energy cost (CNY/kWh)	3000
Unit operation and maintenance cost (CNY/kWh)	0.05
Longevity (Years)	10
Efficiency (dimensionless)	0.9

- Wind Power + Load Power



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- From here they could use this data in their model equation to solve System unit cost of supplying electricity, but they did not.

- To then evaluate the impact of carbon pricing, wind power use, and energy storage integration respectively, the authors created 7 different scenarios to test:

- Scenario 1: All thermal power units are operational, with no wind power or energy storage integration. The carbon capture price is set at 60 CNY per ton.
- Scenario 2: All thermal power units are operational, with no wind power or energy storage integration. The carbon capture price is set at 80 CNY per ton.
- Scenario 3: All thermal power units are operational, with no wind power or energy storage integration. The carbon capture price is set at 100 CNY per ton.
- Scenario 4: A 300 MW wind turbine replaces thermal unit 3 in Scenario 1, while the remaining aspects are consistent with Scenario 1.
- Scenario 5: A 600 MW wind turbine replaces thermal unit 2 in Scenario 1, while the other elements remain unchanged.
- Scenario 6: A 1200 MW wind turbine replaces thermal units 2 and 3 in Scenario 1. The carbon capture price varies between 0, 60, 80, and 100 CNY per ton, while the rest of the scenario remains the same as in Scenario 1.
- Scenario 7: A 1200 MW wind turbine replaces thermal units 2 and 3 in Scenario 1. Energy storage is introduced, and the carbon capture price varies between 0, 60, 80, and 100 CNY per ton. The rest of the elements remain consistent with Scenario 1.

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- Impact of different CC prices

Scenario	Carbon Capture Price (CNY per ton)	Thermal Cost (kCNY)	Carbon Capture Cost (kCNY)	Wind O&M Cost (kCNY)	Wind Curtailment Loss (kCNY)	Load Loss (kCNY)	Energy Storage Capacity (MWh)	Energy Storage Power (MW)	Total Generation Cost (kCNY)	Unit Cost of Supply (CNY/kWh)	CO <sub>2</sub> Emissions (ton)
Baseline scenarios (no wind/storage)											
Scenario 1	-	244.537	68.288	-	-	-	-	-	321.861	0.202	11381.333
Scenario 2	-	244.537	91.05	-	-	-	-	-	335.624	0.216	11381.25
Scenario 3	-	244.537	113.812	-	-	-	-	-	358.386	0.231	11381.2
Wind integration scenarios (no storage)											
Scenario 4	-	201.892	58.939	9.115	0.603	-	-	-	-	0.174	-
Scenario 5	-	167.362	50.906	18.23	8.066	9.843	-	-	-	0.164	-
Energy storage scenarios (with wind integration)											
Scenario 6	0	-	-	-	-	-	0	0	-	0.12276	-
	60	-	-	-	-	-	0	0	-	0.14462	-
	80	-	-	-	-	-	0	0	-	0.15191	-
	100	-	-	-	-	-	0	0	-	0.1592	-
Scenario 7	0	-	-	-	-	-	57.389	44.72	-	0.12295	-
	60	-	-	-	-	-	127.633	40.916	-	0.14501	-
	80	-	-	-	-	-	45.568	74.929	-	0.15209	-
	100	-	-	-	-	-	45.332	19.703	-	0.15934	-

- 
- As CC price increases
  - unit power supply cost increases
  - Total CC cost increases
  - Total power generation cost increases
  - CO<sub>2</sub> emissions decrease
  - Cost of Thermal power stays the same
  - Rises in carbon capture price create a system with positive outcomes for energy conservation and emission reduction, but negative outcomes for cost reduction.
- Increased wind power integration =
  - Cost of thermal power decreases
  - Total CC cost decreases
  - Unit power supply cost decreases
  - Increased wind power integration lowers costs and is more environmentally friendly, leading to a more sustainable energy system. However input and output powers in the system must stay relatively balanced, so increasing wind power too much will cause instability in power supply. (more output than in)
- Impact of Energy Storage
  - Balances out power fluctuations caused by wind power integration, making it more stable and reliable
  - Adds no significant cost increase to unit power supply and is therefore very economically feasible.
  - As CC prices rise, Energy storage is utilized more heavily to reduce high-emission thermal unit use.

Section 6

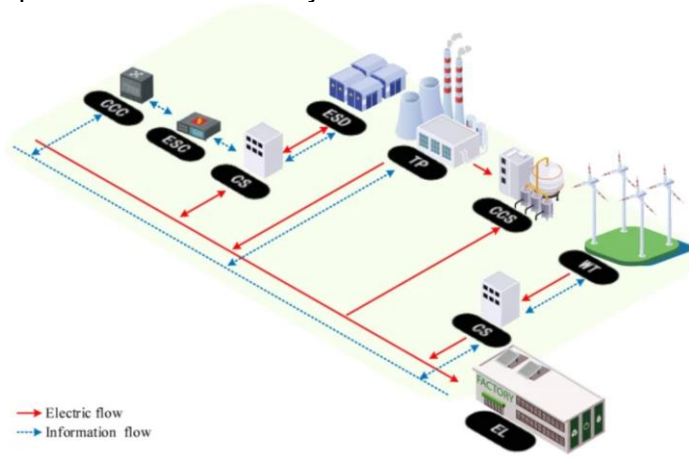
- Successfully developed a “low-carbon optimal scheduling model” that integrated wind power, aiming for energy sustainability, carbon emission reduction, and cost efficiency.
- Key Findings:
  - Wind power decreased costs significantly while maintaining a balanced system with the assistance of ESS.
  - Increasing CC price led to higher total costs but also reduced CO<sub>2</sub> emissions. Showing the trade-off between cost and environmental benefits.
  - Energy Storage Systems enabled the system to use more wind power and be more economically efficient.
- The model will also work under wind deviations or price spikes, due to the adjustable factors included in the model.
- The model showed the importance of a balanced approach involving wind power, carbon capture, and energy storage technologies to create a low-carbon, stable, and cost-effective system.
- Future Work:
  - Use stochastic programming (optimizing a solution to fit multiple outcomes) to quantify the results
  - Incorporate different Renewable Energy, or possibly multiple simultaneously.
  - Update with new Carbon Capture technology?

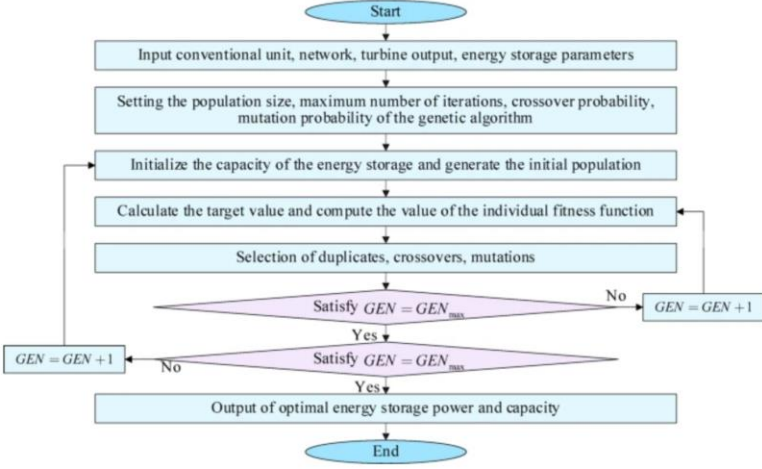
**Research Question/Problem/Need**

Develop a model for optimizing low-carbon levels incorporating post-combustion CC technology, energy storage and an improved GA to optimize efficiently.

**Important Figures**

- Shows all parts involved in the System:



	 <pre> graph TD     Start([Start]) --&gt; Input[Input conventional unit, network, turbine output, energy storage parameters]     Input --&gt; Settings[Setting the population size, maximum number of iterations, crossover probability, mutation probability of the genetic algorithm]     Settings --&gt; Init[Initialize the capacity of the energy storage and generate the initial population]     Init --&gt; Calc[Calculate the target value and compute the value of the individual fitness function]     Calc --&gt; Sel[Selection of duplicates, crossovers, mutations]     Sel --&gt; Dec1{Satisfy GEN = GEN_max}     Dec1 -- No --&gt; Gen1[GEN = GEN + 1]     Gen1 --&gt; Calc     Dec1 -- Yes --&gt; Dec2{Satisfy GEN = GEN_max}     Dec2 -- No --&gt; Gen2[GEN = GEN + 1]     Gen2 --&gt; Calc     Dec2 -- Yes --&gt; Output[Output of optimal energy storage power and capacity]     Output --&gt; End([End])   </pre> <p>• The flow chart of an improved GA algorithm.</p>
<b>VOCAB: (w/definition)</b>	<ul style="list-style-type: none"> <li>• GA       <ul style="list-style-type: none"> <li>○ Genetic Algorithm</li> </ul> </li> <li>• ESS       <ul style="list-style-type: none"> <li>○ Energy Storage System</li> </ul> </li> <li>• CAES       <ul style="list-style-type: none"> <li>○ Compressed Air Energy Storage</li> </ul> </li> <li>• CCS       <ul style="list-style-type: none"> <li>○ Carbon Capture and Storage</li> </ul> </li> <li>• WGANs       <ul style="list-style-type: none"> <li>○ Wasserstein Generative Adversarial Networks</li> </ul> </li> <li>• MWh       <ul style="list-style-type: none"> <li>○ Megawatt per hour</li> </ul> </li> <li>• kWh       <ul style="list-style-type: none"> <li>○ Kilowatt per hour</li> </ul> </li> <li>• CNY       <ul style="list-style-type: none"> <li>○ The carbon price that is paid for every ton of carbon dioxide released into the atmosphere.</li> </ul> </li> </ul>
<b>Cited references to follow up on</b>	<p>Østergaard, P. A., Duic, N., Noorollahi, Y., &amp; Kalogirou, S. (2022). Renewable energy for sustainable development. <i>Renewable Energy</i>, 199, 1145–1152. <a href="https://doi.org/10.1016/j.renene.2022.09.065">https://doi.org/10.1016/j.renene.2022.09.065</a></p> <p>Østergaard, P. A., Duic, N., Noorollahi, Y., &amp; Kalogirou, S. (2023). Advances in renewable energy for sustainable development. <i>Renewable Energy</i>, 219, 119377. <a href="https://doi.org/10.1016/j.renene.2023.119377">https://doi.org/10.1016/j.renene.2023.119377</a></p>
<b>Follow up Questions</b>	<ul style="list-style-type: none"> <li>• How will the model change when incorporating other renewable energy sources?</li> <li>• Could multiple renewable energy sources be combined?</li> <li>• What about multiple Energy Storage Units?</li> </ul>

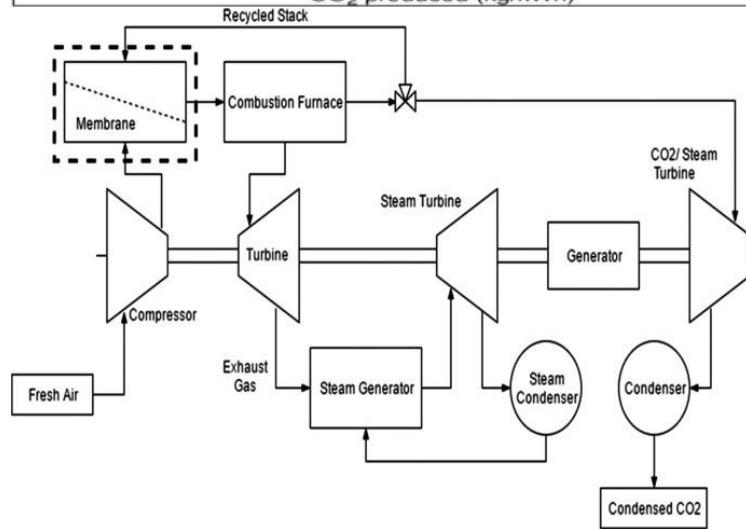
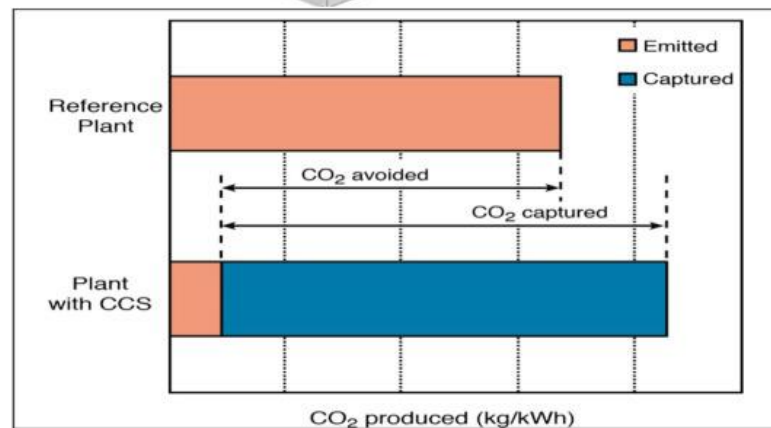
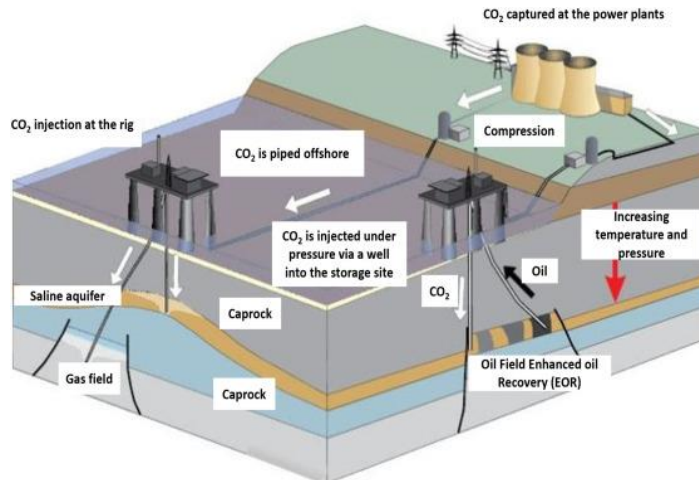
## Article #2 Notes: Progress in carbon capture technologies

Article notes should be on separate sheets

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<b>Source Title</b>	Progress in carbon capture technologies
<b>Source citation (APA Format)</b>	Wilberforce, T., Olabi, A. G., Sayed, E. T., Elsaid, K., & Abdelkareem, M. A. (2021). Progress in carbon capture technologies. <i>Science of The Total Environment</i> , 761, 143203. <a href="https://doi.org/10.1016/j.scitotenv.2020.143203">https://doi.org/10.1016/j.scitotenv.2020.143203</a>
<b>Original URL</b>	<a href="https://doi.org/10.1016/j.scitotenv.2020.143203">https://doi.org/10.1016/j.scitotenv.2020.143203</a>
<b>Source type</b>	Scientific Article
<b>Keywords</b>	Carbon Capture; Combustion; Transportation; Diffusion
<b>#Tags</b>	#Current Progress
<b>Summary of key points + notes (include methodology)</b>	<p>Human-related factors are the key contributors to carbon dioxide emissions.</p> <p>Carbon Capture and storage are reliable tools that can address this global challenge</p> <p>CCS is the separation of CO<sub>2</sub> to a regulated location for ideal storage.</p> <p>We continue to rely on fossil energy and therefore CCS is critical.</p> <p>Most CCS technologies can absorb 85-95% of CO<sub>2</sub>, but at the cost of 10-40% more energy.</p> <p>The use and advancement of CCS is limited by the cost and energy required to utilize it.</p>
<b>Research Question/Problem/Need</b>	What current Carbon Capture technologies are available, and what are the advantages & disadvantages of each?

Important Figures



VOCAB: (w/definition)

- CCS
  - Carbon Capture & Storage
- PSA
  - Pressure Swing Absorber
- WGS
  - Water Gas Shift

	<ul style="list-style-type: none"> <li>• MOF <ul style="list-style-type: none"> <li>○ Metal Organic Framework</li> </ul> </li> </ul>
<b>Cited references to follow up on</b>	<p>Abdelkareem, M. A., El Haj Assad, M., Sayed, E. T., &amp; Soudan, B. (2018). Recent progress in the use of renewable energy sources to power water desalination plants. <i>Desalination</i>, 435, 97–113.  <a href="https://doi.org/10.1016/j.desal.2017.11.018">https://doi.org/10.1016/j.desal.2017.11.018</a></p> <p>Abdelkareem, M. A., Elsaid, K., Wilberforce, T., Kamil, M., Sayed, E. T., &amp; Olabi, A. (2021). Environmental aspects of fuel cells: A review. <i>Science of The Total Environment</i>, 752, 141803.  <a href="https://doi.org/10.1016/j.scitotenv.2020.141803">https://doi.org/10.1016/j.scitotenv.2020.141803</a></p>
<b>Follow up Questions</b>	<ul style="list-style-type: none"> <li>• Where is Carbon Capture Technology now?</li> <li>• What is the cheapest Carbon Capture Technology?</li> <li>• What is the most effective Carbon Capture Technology?</li> </ul>

## Article #3 Notes: Carbon capture technology to produce clean fuel from air

Article notes should be on separate sheets

### KEEP THIS BLANK AND USE AS A TEMPLATE

<b>Source Title</b>	Carbon capture technology to produce clean fuel from air
<b>Source citation (APA Format)</b>	University of Surrey. (2025, April 3). Carbon capture technology to produce clean fuel from air. <i>ScienceDaily</i> . Retrieved October 9, 2025, from <a href="http://www.sciencedaily.com/releases/2025/04/250403122938.htm">www.sciencedaily.com/releases/2025/04/250403122938.htm</a>
<b>Original URL</b>	<a href="https://www.sciencedaily.com/releases/2025/04/250403122938.htm">https://www.sciencedaily.com/releases/2025/04/250403122938.htm</a>
<b>Source type</b>	Science Journal
<b>Keywords</b>	Carbon Capture; DFM; DAC; Synthetic Fuel
<b>#Tags</b>	#Synthetic Fuel
<b>Summary of key points + notes (include methodology)</b>	<ul style="list-style-type: none"> <li>• Unique carbon capture technology offers a more cost-effective way to turn CO<sub>2</sub> from the air into clean synthetic fuel.</li> </ul>

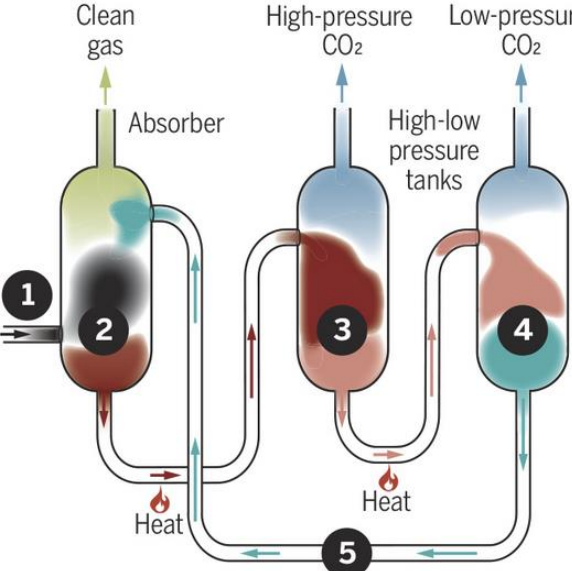
	<ul style="list-style-type: none"> <li>• Dual-Function Material is a process that combines carbon capture and conversion. It uses green hydrogen from renewable electricity and carbon from the atmosphere to create fuel. This fuel could allow industries to effectively have zero net emissions.</li> <li>• DFM is capable of removing carbon at a cost of below \$400 per ton, but currently operates at \$740 per ton. DFM holds promise for future large-scale deployment.</li> </ul>
<b>Research Question/Problem/Need</b>	Current Carbon Capture technologies are too expensive and offer little in the way of converting it back into clean usable energy.
<b>Important Figures</b>	None given
<b>VOCAB: (w/definition)</b>	<ul style="list-style-type: none"> <li>• DFM <ul style="list-style-type: none"> <li>○ Dual-Function Material</li> </ul> </li> <li>• DAC <ul style="list-style-type: none"> <li>○ Direct Air Capture</li> </ul> </li> <li>• Superstructure optimization <ul style="list-style-type: none"> <li>○ An advanced modeling technique</li> </ul> </li> </ul>
<b>Cited references to follow up on</b>	Meshkat Dolat, Kamran Keynejad, Melis S. Duyar, Michael Short. Superstructure optimization of direct air capture integrated with synthetic natural gas production. <i>Applied Energy</i> , 2025; 384: 125413 DOI: <a href="https://doi.org/10.1016/j.apenergy.2025.125413">10.1016/j.apenergy.2025.125413</a>
<b>Follow up Questions</b>	<ul style="list-style-type: none"> <li>• Are there other energy sources that would function with this process? Maybe solar or wind energy?</li> </ul>

# Article #4 Notes: Carbon Capture Marches Toward Practical Use

Article notes should be on separate sheets

## KEEP THIS BLANK AND USE AS A TEMPLATE

<b>Source Title</b>	Carbon Capture Marches Toward Practical Use
<b>Source citation (APA Format)</b>	Service, R. F. (2021). Carbon capture marches toward practical use. <i>Science</i> , 371(6536), 1300–1300. <a href="https://doi.org/10.1126/science.371.6536.1300">https://doi.org/10.1126/science.371.6536.1300</a>
<b>Original URL</b>	<a href="https://doi.org/10.1126/science.371.6536.1300">https://doi.org/10.1126/science.371.6536.1300</a>
<b>Source type</b>	Science Article
<b>Keywords</b>	Amines; Molecules; Carbon Capture; Organic Water Solvent
<b>#Tags</b>	#Amines #Water Solvent
<b>Summary of key points + notes (include methodology)</b>	<p>Carbon Capture is necessary to “Stave off the worst of Climate Change”</p> <p>Most popular form of carbon capture is too expensive</p> <p>New chemical CO<sub>2</sub> trap reduces cost by 20%</p> <p>Nearing Commercial Viability</p> <p>Amines dissolved in water grab CO<sub>2</sub> in the air, but must be boiled and recondensed to reuse (increasing energy and cost)</p> <p>In order to increase to necessary levels, carbon capture must drop from \$58 per ton collected, to \$30 per ton.</p> <p>PPNL created liquid organic solvents, containing CO<sub>2</sub> grabbing amines with no need to be boiled or recondensed.</p> <p>Unsuccessful at first, so they tweaked it creating a molecule called 2-EEMPA which worked much better.</p> <p>2-EEMPA requires %17 less energy and dropped the cost to \$47 per ton.</p>

	2-EEMPA has inspired other researchers to make their own solvents, and has garnered support from the government (\$4.9 billion for carbon capture projects!)
<b>Research Question/Problem/ Need</b>	Create a more efficient organic water solvent for Carbon Capture to save energy and money.
<b>Important Figures</b>	 <p>The diagram illustrates a cyclic process for carbon capture using 2-EEMPA. It consists of three main vessels: an Absorber, a High-pressure tank, and a Low-pressure tank, connected by pipes. The process is numbered 1 through 5:</p> <ol style="list-style-type: none"> <li><b>1</b> CO<sub>2</sub>-rich flue gas (●) from the power plant enters an absorber vessel.</li> <li><b>2</b> Organic solvent absorbs CO<sub>2</sub> (●) and releases clean exhaust to the atmosphere (●).</li> <li><b>3</b> Solvent rich in CO<sub>2</sub> is heated and piped to a high-pressure tank, where CO<sub>2</sub> (●) is released and piped away for storage.</li> <li><b>4</b> Solvent with less CO<sub>2</sub> (●) is heated and piped to a low-pressure tank to remove the remaining CO<sub>2</sub>.</li> <li><b>5</b> Cleaned solvent (●) is returned to the absorber for reuse.</li> </ol> <p>Labels in the diagram include: Clean gas, Absorber, High-pressure CO<sub>2</sub>, Low-pressure CO<sub>2</sub>, High-low pressure tanks, Heat, and numbered steps 1-5.</p>
<b>VOCAB: (w/definition)</b>	<ul style="list-style-type: none"> <li>• 2-EEMPA <ul style="list-style-type: none"> <li>○ Name of solvent created by authors</li> </ul> </li> </ul>
<b>Cited references to follow up on</b>	Gensheng Li, Jinsheng Sun, Zhangxin Chen, Zhenhua Rui. Editorial for the Special Issue on Carbon Capture, Utilization, and Storage. <i>Engineering</i> , 2025, 48(5): 1-2 DOI:10.1016/j.eng.2025.04.004
<b>Follow up Questions</b>	<p>In what ways could this solvent be improved?</p> <p>How long/what will it take to make this commercially viable?</p>

## Article #5 Notes: Polymeric membranes in carbon capture, utilization, and storage: current trends and future directions in decarbonization of industrial flue gas and climate change mitigation

<b>Source Title</b>	Polymeric membranes in carbon capture, utilization, and storage: current trends and future directions in decarbonization of industrial flue gas and climate change mitigation
<b>Source citation (APA Format)</b>	Mollahosseini, A., Nikkiah Dafchahi, M., Khoshhal Salestan, S., Chew, J. W., Mozafari, M., Soroush, M., Hrapovic, S., Hemraz, U. D., Giro, R., Steiner, M. B., La, Y.-H., Seyedpour Taji, S. F., Azyat, K., Islam, M. A., Kavyani, S., Wang, X., Cho, J.-Y., & Sadrzadeh, M. (2025). Polymeric membranes in carbon capture, utilization, and storage: Current trends and future directions in decarbonization of industrial flue gas and climate change mitigation. <i>Energy &amp; Environmental Science</i> , 18(11), 5025–5092. <a href="https://doi.org/10.1039/D4EE05328A">https://doi.org/10.1039/D4EE05328A</a>
<b>Original URL</b>	<a href="https://doi.org/10.1039/D4EE05328A">https://doi.org/10.1039/D4EE05328A</a>
<b>Source type</b>	Scientific Article
<b>Keywords</b>	polymeric membranes; flue gas; CO <sub>2</sub> /N <sub>2</sub> separation; process simulation; carbon utilization; computational design
<b>#Tags</b>	#Membranes #FlueGas
<b>Summary of key points + notes (include methodology)</b>	<p>This review surveys recent advances in polymeric membranes designed for carbon capture from industrial flue gas streams. It covers materials, membrane module design, separation performance (e.g. permeance and selectivity), stability, and integration with broader CCUS (Carbon Capture, Utilization, and Storage) systems.</p> <p>The paper emphasizes the role of computational methods—including AI—to accelerate materials discovery, structure-performance prediction, and optimization of membrane-modifier combinations.</p> <p>It discusses challenges: dealing with impurities in flue gas (SO<sub>x</sub>, NO<sub>x</sub>, moisture), tradeoff between permeability (how much CO<sub>2</sub> gets through quickly) and selectivity (how pure that separated CO<sub>2</sub> stream is), and membrane long-term stability (mechanical, thermal).</p>

	Methodologically, the review draws from lab experiments, pilot-scale tests, process simulations, and techno-economic comparisons across membrane technologies.
<b>Research Question/Problem/ Need</b>	How can polymeric membrane technologies be improved (in materials, module design, and integration) to reach high performance, long durability, and economic viability for capturing CO <sub>2</sub> from industrial flue gas under real operating conditions?
<b>Important Figures</b>	<p>Fuel is processed to produce H<sub>2</sub> with CO<sub>2</sub> as a side product</p> <p>Separation of CO<sub>2</sub> happens after the process</p> <p>Separation of CO<sub>2</sub> happens after the process</p>
<b>VOCAB: (w/definition)</b>	<ul style="list-style-type: none"> <li>• <b>Permeance</b> — measure of how much gas passes through a membrane per unit time, area, and pressure difference.</li> <li>• <b>Selectivity</b> — ratio of how well a membrane separates CO<sub>2</sub> relative to another gas (often N<sub>2</sub> in flue gas).</li> <li>• <b>Membrane module</b> — the hardware setup of membrane sheets or fibers, housing, flow channels, etc., through which gas passes.</li> <li>• <b>Computational materials design</b> — using simulations, AI, and modelling to predict how material structure and composition will affect performance, so fewer trial-and-error lab tests are needed.</li> </ul>
<b>Cited references to follow up on</b>	<ul style="list-style-type: none"> <li>• Studies of pilot-scale polymeric membrane modules under real flue gas conditions (with impurities) to assess durability.</li> </ul>
<b>Follow up Questions</b>	<ul style="list-style-type: none"> <li>• What other impurities in flue gas are apparent?</li> </ul>

## Article #6 Notes: A Review of Materials for Carbon Dioxide Capture

<b>Source Title</b>	A Review of Materials for Carbon Dioxide Capture
<b>Source citation (APA Format)</b>	School for Engineering of Matter, Transport, and Energy, Arizona State University. (2025). <i>A Review of Materials for Carbon Dioxide Capture</i> . <i>Catalysts</i> , 15(3), 273. <a href="https://doi.org/10.3390/catal15030273">https://doi.org/10.3390/catal15030273</a>
<b>Original URL</b>	<a href="https://doi.org/10.3390/catal15030273">https://doi.org/10.3390/catal15030273</a>
<b>Source type</b>	Review Article
<b>Keywords</b>	CO <sub>2</sub> capture; sorbents; materials design; adsorption; industrial applications
<b>#Tags</b>	#Materials #Sorbents #CO2Capture
<b>Summary of key points + notes (include methodology)</b>	<p>The article surveys recent advancements in materials used for CO<sub>2</sub> capture, including adsorbents, membranes, and hybrid systems.</p> <p>It describes performance metrics (capacity, selectivity, regeneration energy) and highlights gaps in scalability and durability of many lab-scale materials.</p> <p>Methodology: meta-analysis of published performance data, classification of materials by type and capture scenario (post-combustion, pre-combustion, direct air capture).</p> <p>The review identifies key research directions: reducing regeneration energy, improving cycle stability, and bridging lab performance to real-world conditions (impurities, moisture, large scale).</p>
<b>Research Question/Problem/ Need</b>	Which next-generation sorbent or material systems can both achieve high capture performance and meet practical durability, cost, and operational conditions for industrial CO <sub>2</sub> capture?

<p><b>Important Figures</b></p>	
<p><b>VOCAB: (w/definition)</b></p>	<p>Regeneration energy — the energy required to restore a sorbent to its original state so it can capture CO<sub>2</sub> again.</p> <p>Selectivity — the ability of a material to preferentially capture CO<sub>2</sub> over other gases (e.g., N<sub>2</sub> in flue gas).</p>
<p><b>Cited references to follow up on</b></p>	<p>Zhu, Q., &amp; Wang, X. (2024). Durable adsorbents for carbon capture: bridging lab and industry. <i>Journal of CO<sub>2</sub> Utilization</i>, 60, 102056. <a href="https://doi.org/10.1016/j.jcou.2024.102056">https://doi.org/10.1016/j.jcou.2024.102056</a></p>
<p><b>Follow up Questions</b></p>	<p>What are the main degradation mechanisms for high-capacity sorbents when exposed to real flue-gas impurities over thousands of cycles?</p> <p>How do the energy penalties of different material regeneration methods compare when scaled to a full-scale industrial CO<sub>2</sub> capture plant?</p> <p>What cost thresholds (\$/ton CO<sub>2</sub> captured) must next-generation materials meet to be commercially viable in major industries like cement or steel?</p>

## Article #7 Notes: Recent advancements in carbon capture materials research: Innovative optimization of materials synthesis and engineering applications

<b>Source Title</b>	Recent advancements in carbon capture materials research: Innovative optimization of materials synthesis and engineering applications
<b>Source citation (APA Format)</b>	Wang, Y., Gao, F., Niu, Y., Chen, K., Yi, H., Zhang, J., Zhou, Y., Tang, X., & Zhao, S. (2025). Recent advancements in carbon capture materials research: Innovative optimization of materials synthesis and engineering applications. <i>Journal of Materials Chemistry A</i> , 13, 23323-23353. <a href="https://doi.org/10.1039/D5TA01304F">https://doi.org/10.1039/D5TA01304F</a>
<b>Original URL</b>	<a href="https://doi.org/10.1039/D5TA01304F">https://doi.org/10.1039/D5TA01304F</a>
<b>Source type</b>	Review Article
<b>Keywords</b>	MOFs; porous polymers; adsorbent engineering; material synthesis; scale-up
<b>#Tags</b>	#MOFs #Adsorbents
<b>Summary of key points + notes (include methodology)</b>	<p>This review focuses on material-level innovations: metal–organic frameworks (MOFs), porous organic polymers (POPs), functional carbon-based adsorbents, and composites engineered for CO<sub>2</sub> capture under realistic conditions.</p> <p>It emphasizes structural optimization (pore size, surface chemistry), advanced synthesis methods (microwave, solvent-free, templating), and engineering applications (module design, process embedding).</p> <p>Methodology: systematic survey of literature between circa 2022-2025, classification of advancements by material type and performance improvements, highlighting “engineering readiness”.</p> <p>Challenges noted: cost of synthesis, reproducibility, integration into modules, and moving from bench to pilot scale.</p>
<b>Research Question/Problem/Need</b>	How can advanced materials (MOFs, POPs, composites) be synthesized and engineered in ways that maintain high performance while being cost-effective and scalable for industrial CO <sub>2</sub> capture?

<b>Important Figures</b>	
<b>VOCAB: (w/definition)</b>	<p>MOFs — Metal–Organic Frameworks: a type of porous crystalline material constructed from metal nodes and organic linkers, with high surface area and tunable chemistry.</p> <p>POPs — Porous Organic Polymers: amorphous polymeric materials with controlled porosity used for gas adsorption.</p>
<b>Cited references to follow up on</b>	<p>Smith, A. J., &amp; Lee, M. (2023). Cost modeling of MOF synthesis for carbon capture applications. <i>ACS Applied Materials &amp; Interfaces</i>, 15(4), 2345-2357. <a href="https://doi.org/10.1021/acsami.2c10382">https://doi.org/10.1021/acsami.2c10382</a></p> <p>Chen, Y., et al. (2024). Scale-up and module-integration of porous polymer adsorbents: from lab to pilot. <i>Industrial &amp; Engineering Chemistry Research</i>, 63(12), 4489-4503. <a href="https://doi.org/10.1021/ie403345h">https://doi.org/10.1021/ie403345h</a></p>
<b>Follow up Questions</b>	<p>What synthesis methods show the most promise for reducing cost of advanced adsorbents while retaining high CO<sub>2</sub> capture performance?</p> <p>How does the performance of material modules (not just powders) degrade when exposed to real-world flue gas (moisture, SO<sub>x</sub>, NO<sub>x</sub>) over time?</p> <p>What are the key bottlenecks in bundling advanced adsorbent materials into full-scale capture systems (including system engineering, lifetime, maintenance)?</p>

## Article #8 Notes: A critical appraisal of advances in integrated CO<sub>2</sub> capture and electrochemical conversion

<b>Source Title</b>	A critical appraisal of advances in integrated CO <sub>2</sub> capture and electrochemical conversion
<b>Source citation (APA Format)</b>	Badreldin, A., & Li, Y. (2025). A critical appraisal of advances in integrated CO <sub>2</sub>

	capture and electrochemical conversion. Chemical Science, 16, 2483-2513. <a href="https://doi.org/10.1039/D4SC06642A">https://doi.org/10.1039/D4SC06642A</a>
Original URL	<a href="https://pubs.rsc.org/en/content/articlelanding/2025/sc/d4sc06642a">https://pubs.rsc.org/en/content/articlelanding/2025/sc/d4sc06642a</a>
Source type	Scientific Article
Keywords	Electrochemical CO <sub>2</sub> capture; reactive capture; adsorptive capture; conversion to value-added products
#Tags	#ElectrochemicalCCS #Conversion #ReactiveCapture
Summary of key points + notes (include methodology)	<p>The paper examines integrated processes where CO<sub>2</sub> capture and conversion are combined (rather than separate). Two routes: (1) electrochemical reactive capture (e.g., amine/bicarbonate mediated) and (2) adsorptive capture &amp; conversion (ACC) from flue gas.</p> <p>The authors highlight key technological metrics (energy input per mole CO<sub>2</sub>, Faradaic efficiencies, catalyst durability) and compare these to conventional sequential capture + conversion systems.</p> <p>Methodology: review of recent literature (2020-2024), benchmarking performance of emerging systems, identifying techno-economic and scaling challenges (materials, system integration).</p> <p>The article points to major gaps: system stability, separation of product and CO<sub>2</sub> stream purity, realistic cost modelling and lifecycle assessment.</p>
Research Question/Problem/Need	How can integrated capture-and-conversion systems be engineered to reduce overall energy and cost penalties and achieve commercial readiness, while maintaining durability and product quality?
Important Figures	<p>The figure consists of three process flow diagrams labeled a, b, and c, illustrating integrated CO<sub>2</sub> capture and electrochemical conversion systems.</p> <ul style="list-style-type: none"> <li><b>Diagram a:</b> Shows a system starting with a Point Source of CO<sub>2</sub> entering a CO<sub>2</sub> Free Gas stream. This stream goes through a CO<sub>2</sub> Compression unit and a Stripper. The compressed CO<sub>2</sub> is then fed into an Electrolyzer Stack (Cathode and Anode). The Electrolyzer Stack produces Purified O<sub>2</sub> and Recycled CO<sub>2</sub>. The Recycled CO<sub>2</sub> is fed into two Catholyte PSA No. 1 and No. 2 units. The PSA units produce Hydrocarbon (g) and Recycled CO<sub>2</sub>. The Electrolyzer Stack also produces Anolyte Recycled CO<sub>2</sub>, which is fed into an Anolyte PSA for CO<sub>2</sub> Recycle from Carbonate Crossover. The PSA units produce Overhead Gases and Liquid Products (i.e., HCOOH, CH<sub>3</sub>OOH).</li> <li><b>Diagram b:</b> Shows a system starting with a Point Source of CO<sub>2</sub> entering a CO<sub>2</sub> Free Gas stream. This stream goes through a KOH Contactor Tower. The CO<sub>2</sub> Free Gas is then fed into a Rich Amine or (Bi)Carbonate unit. The Rich Amine or (Bi)Carbonate is then fed into a PSA for Hydrocarbon Gas Purification and CO<sub>2</sub> Separation. The PSA produces Purified Hydrocarbon Gas Product (i.e., CO, C<sub>2</sub>H<sub>4</sub>) and Recycled CO<sub>2</sub>. The Recycled CO<sub>2</sub> is fed into an Electrolyzer Stack (Cathode and Anode). The Electrolyzer Stack produces Purified O<sub>2</sub> and Vapors of Liquid Products (i.e., HCOOH, CH<sub>3</sub>OOH). The Vapors of Liquid Products are then fed into Distillation Columns. The Distillation Columns produce Lean Amine or (Bi)Carbonate, which is recycled back to the Rich Amine or (Bi)Carbonate unit.</li> <li><b>Diagram c:</b> Shows a system starting with a Point Source of CO<sub>2</sub> entering a CO<sub>2</sub> Free Gas stream. This stream goes through a CO<sub>2</sub> Compression unit and a Stripper. The compressed CO<sub>2</sub> is then fed into an Electrolyzer Stack (Cathode and Anode). The Electrolyzer Stack produces Purified O<sub>2</sub> and Recycled CO<sub>2</sub>. The Recycled CO<sub>2</sub> is fed into two Catholyte PSA No. 1 and No. 2 units. The PSA units produce Hydrocarbon (g) and Recycled CO<sub>2</sub>. The Electrolyzer Stack also produces Anolyte Recycled CO<sub>2</sub>, which is fed into an Anolyte PSA for CO<sub>2</sub> Recycle from Carbonate Crossover. The PSA units produce Overhead Gases and Liquid Products (i.e., HCOOH, CH<sub>3</sub>OOH).</li> </ul>

<b>VOCAB: (w/definition)</b>	<p>Faradaic efficiency — fraction of electric charge that accomplishes the desired chemical reaction (here CO<sub>2</sub> reduction) rather than side reactions.</p> <p>Reactive capture — process where CO<sub>2</sub> is captured and immediately converted in one step or tightly coupled step, reducing intermediate handling.</p>
<b>Cited references to follow up on</b>	<p>Wang, H., &amp; Zhang, S. (2023). Integrated capture and electrochemical conversion of CO<sub>2</sub>: system modelling and optimization. <i>Energy &amp; Environmental Science</i>, 16(11), 5496-5512.  <a href="https://doi.org/10.1039/D3EE02060">https://doi.org/10.1039/D3EE02060</a></p>
<b>Follow up Questions</b>	<p>Which capture + conversion pathway (eRCC or ACC) offers the lowest projected cost per tonne of CO<sub>2</sub> removed and useful product generated?</p> <p>What are the key durability and maintenance issues in integrated capture-conversion cells under industrial flue-gas conditions?</p> <p>How realistic are current techno-economic projections for value-added product generation from captured CO<sub>2</sub> (e.g., fuels, chemicals) at scale?</p>

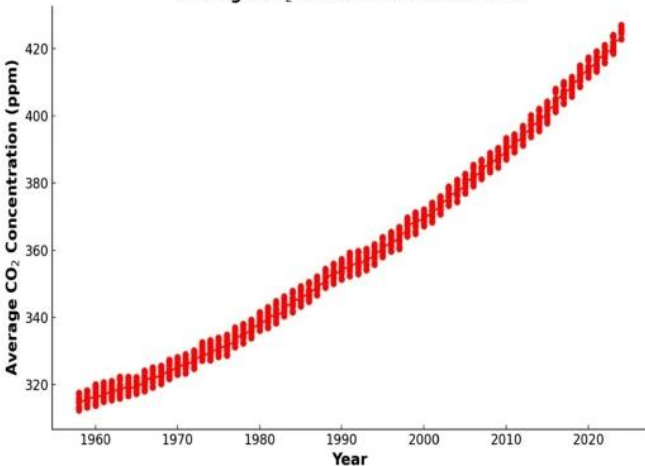
## Article #9 Notes: Carbon Capture by Biological Methods

<b>Source Title</b>	Carbon Capture by Biological Methods
<b>Source citation (APA Format)</b>	<p>Fu, W., Liu, Y., &amp; et al. (2025). Carbon capture by biological methods. <i>Cambridge Prisms: Carbon Technologies</i>, 1, e5.  <a href="https://doi.org/10.1017/cat.2025.10005">https://doi.org/10.1017/cat.2025.10005</a></p>
<b>Original URL</b>	<p><a href="https://www.cambridge.org/core/journals/cambridge-prisms-carbon-technologies/article/carbon-capture-by-biological-methods/2F7142CC35A605AC298301253E437F26">https://www.cambridge.org/core/journals/cambridge-prisms-carbon-technologies/article/carbon-capture-by-biological-methods/2F7142CC35A605AC298301253E437F26</a></p>
<b>Source type</b>	Review Article
<b>Keywords</b>	Bio-capture; algae; microorganisms; biofilm; carbon uptake
<b>#Tags</b>	#BiologicalCCS #BioCapture #Microorganisms
<b>Summary of key points + notes (include methodology)</b>	<p>This review covers carbon capture via biological methods: plants, algae, microbes, biofilms, and their potential role in industrial CO<sub>2</sub> capture and utilization.</p> <p>It discusses natural pathways (photosynthesis, biofilm formation), engineered systems (microbial electrosynthesis, algae bioreactors capturing flue CO<sub>2</sub>), and their real-world constraints (land/space, growth rates, contamination,</p>

	<p>scalability).</p> <p>Methodology: literature review of recent (2020-2024) biological capture systems, analysis of performance metrics (capture rate per area, cost, energy input), and discussion of integration into industrial or waste-gas streams.</p>
<b>Research Question/Problem/Need</b>	What biological systems (organisms, bioreactors) can capture CO <sub>2</sub> at sufficient rates, cost-effectively and sustainably, to contribute meaningfully to industrial carbon removal efforts?
<b>Important Figures</b>	Graphs of CO <sub>2</sub> uptake rate by algae vs land area; comparison of bio-capture vs engineered sorbents in terms of energy and land footprint. (image didn't paste)
<b>VOCAB: (w/definition)</b>	<p>Biofilm — a thin, often microscopic layer of microorganisms that adhere to surfaces and form communities, sometimes used for CO<sub>2</sub> uptake or conversion.</p> <p>Microbial electrosynthesis — using microorganisms to convert CO<sub>2</sub> (often with electricity) into chemicals or fuels.</p>
<b>Cited references to follow up on</b>	<p>Li, J., et al. (2024). Biofilm-based CO<sub>2</sub> conversion to methane: advances and challenges. <i>Energy Conversion and Management</i>, 285, 116682. <a href="https://doi.org/10.1016/j.enconman.2023.116682">https://doi.org/10.1016/j.enconman.2023.116682</a></p> <p>Smith, G., &amp; Thompson, R. (2023). Algal bioreactors for industrial flue-gas CO<sub>2</sub> capture: operational performance and cost-analysis. <i>Bioresource Technology</i>, 367, 128948. <a href="https://doi.org/10.1016/j.biortech.2023.128948">https://doi.org/10.1016/j.biortech.2023.128948</a></p>
<b>Follow up Questions</b>	<p>How does the land/space requirement of biological capture systems compare to conventional sorbent or membrane-based systems for the same tonnage of CO<sub>2</sub> removed?</p> <p>What are the lifecycle environmental impacts (e.g., nutrients, water, energy) of large-scale biological CO<sub>2</sub> capture systems?</p> <p>Can biological capture systems be integrated with industrial waste streams (e.g., cement, steel) and still maintain performance under flue-gas contaminant conditions?</p>

## Article #10 Notes: Carbon Capture, Utilization, and Storage in the MENA Region: A Regional Review of Projects and Challenges

<b>Source Title</b>	Carbon Capture, Utilization, and Storage in the MENA Region: A Regional Review of Projects and Challenges
<b>Source citation (APA Format)</b>	Al-Subhan, K., Haddad, M., & et al. (2025). Carbon Capture, Utilization, and Storage in the MENA Region: A Regional Review of Projects and Challenges. <i>Arabian Journal for Science and Engineering</i> , 50, 4529-4549. <a href="https://doi.org/10.1007/s13369-025-09999-7">https://doi.org/10.1007/s13369-025-09999-7</a>
<b>Original URL</b>	<a href="https://link.springer.com/article/10.1007/s13369-025-09999-7">https://link.springer.com/article/10.1007/s13369-025-09999-7</a>
<b>Source type</b>	Review Article
<b>Keywords</b>	CCUS; Middle East and North Africa; oil & gas industry; policy & infrastructure; CO <sub>2</sub> utilization
<b>#Tags</b>	#CCUS #MENA #RegionalReview
<b>Summary of key points + notes (include methodology)</b>	<p>This article reviews CCUS (carbon capture, utilization, storage) projects and potential in the Middle East &amp; North Africa (MENA) region, focusing on oil &amp; gas sectors, emerging renewables, and regional policies.</p> <p>It examines technical readiness of capture technologies, CO<sub>2</sub> storage potential in local geology, utilization pathways (e.g., enhanced oil recovery, chemical feedstocks), and key obstacles (regulations, cost, infrastructure).</p> <p>Methodology: survey of regional project databases, policy and regulatory frameworks, infrastructure and geology assessments, and comparison with global benchmarks.</p>
<b>Research Question/Problem/ Need</b>	How can CCUS technologies be adapted and scaled in the MENA region, given unique technical, economic, and policy landscapes, to maximize CO <sub>2</sub> mitigation and utilization benefits?

<b>Important Figures</b>	<p><b>Fig. 1</b></p> <p><b>Average CO<sub>2</sub> Concentration over time</b></p>  <p>Average CO<sub>2</sub> concentration in ppm over time (Data obtained from [8])</p>
<b>VOCAB: (w/definition)</b>	<p>CCUS — Carbon Capture, Utilization &amp; Storage: capturing CO<sub>2</sub>, using it (e.g., in chemicals, fuels), and storing it (usually underground) to reduce net emissions.</p> <p>EOR — Enhanced Oil Recovery: injecting CO<sub>2</sub> into oil fields to increase extraction; also a utilization pathway for CO<sub>2</sub>.</p>
<b>Cited references to follow up on</b>	<p>Al-Kaabi, A., &amp; Kumar, P. (2024). Storage potential of CO<sub>2</sub> in Gulf region basalts: modelling and field data. <i>International Journal of Greenhouse Gas Control</i>, 132, 104710. <a href="https://doi.org/10.1016/j.ijggc.2024.104710">https://doi.org/10.1016/j.ijggc.2024.104710</a></p> <p>Haddad, M., et al. (2023). Legal and regulatory frameworks for CCUS deployment in the Middle East: a comparative study. <i>Energy Policy</i>, 178, 113730. <a href="https://doi.org/10.1016/j.enpol.2023.113730">https://doi.org/10.1016/j.enpol.2023.113730</a></p>
<b>Follow up Questions</b>	<p>Which geological formations in the MENA region offer the highest capacity and lowest risk for long-term CO<sub>2</sub> storage, and what are the monitoring/verification costs?</p> <p>How can captured CO<sub>2</sub> be most effectively utilized in MENA economies (e.g., in petrochemicals) and what business models support that?</p> <p>What policy or regulatory gaps in the region are most critical to address for large-scale CCUS deployment (e.g., liability, permanence, transport infrastructure)?</p>

# Article #11 Notes: CO<sub>2</sub> Capture: A Comprehensive Review and Bibliometric Analysis of Scalable Materials and Sustainable Solutions

<b>Source Title</b>	CO <sub>2</sub> Capture: A Comprehensive Review and Bibliometric Analysis of Scalable Materials and Sustainable Solutions
<b>Source citation (APA Format)</b>	Carrascal-Hernández, D. C., Grande-Tovar, C. D., Mendez-Lopez, M., Insuasty, D., García-Freites, S., Sanjuan, M., & Márquez, E. (2025). CO <sub>2</sub> Capture: A Comprehensive Review and Bibliometric Analysis of Scalable Materials and Sustainable Solutions. <i>Molecules</i> , 30(3), 563. <a href="https://www.mdpi.com/1420-3049/30/3/563">https://www.mdpi.com/1420-3049/30/3/563</a>
<b>Original URL</b>	<a href="https://www.mdpi.com/1420-3049/30/3/563">https://www.mdpi.com/1420-3049/30/3/563</a>
<b>Source type</b>	Review article with bibliometric analysis
<b>Keywords</b>	CO <sub>2</sub> capture; scalable materials; bibliometric trends; sustainability; market readiness
<b>#Tags</b>	#ScalableMaterials #Bibliometrics #SustainableCCS
<b>Summary of key points + notes (include methodology)</b>	<p>This paper provides a bibliometric analysis of CO<sub>2</sub> capture research (material trends, citation patterns, funding sources) and couples it with a review of scalable materials (adsorbents, membranes, solvents).</p> <p>It highlights which material classes dominate research attention, which remain under-studied, and how research is shifting toward scalable, lower-cost solutions.</p> <p>Methodology: bibliometric data mining of publications 2000-2024, grouping by material type, country/institution, and forward-looking summary of sustainable solutions and material readiness levels.</p>
<b>Research Question/Problem/Need</b>	What material types (and associated technologies) show the strongest indicators of scaling potential and sustainability in the field of CO <sub>2</sub> capture, and how can research priorities be aligned accordingly?



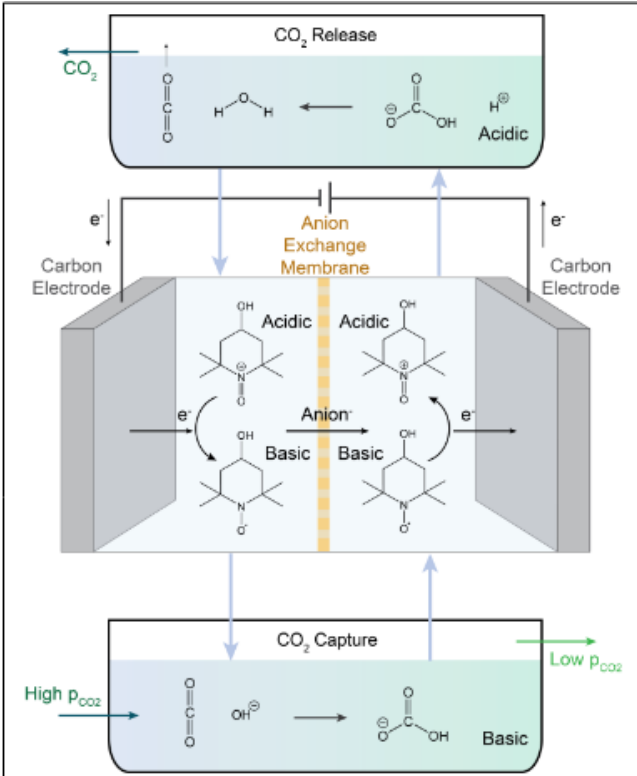
## Article #12 Notes: An ecosystem of carbon dioxide removal reviews – part 1: direct air CO<sub>2</sub> capture and storage

<b>Source Title</b>	An ecosystem of carbon dioxide removal reviews – part 1: direct air CO <sub>2</sub> capture and storage
<b>Source citation (APA Format)</b>	van der Spek, M., Bardow, A., Baum, C. M., Bolongaro, V., Dufour-Décieux, V., Esch, C., ... Minx, J. C. (2025). An ecosystem of carbon dioxide removal reviews – part 1: direct air CO <sub>2</sub> capture and storage. <i>Energy &amp; Environmental Science</i> , 18, 9713–9785. <a href="https://pubs.rsc.org/en/content/articlelanding/2025/ee/d5ee01732g">https://pubs.rsc.org/en/content/articlelanding/2025/ee/d5ee01732g</a>
<b>Original URL</b>	<a href="https://pubs.rsc.org/en/content/articlelanding/2025/ee/d5ee01732g">https://pubs.rsc.org/en/content/articlelanding/2025/ee/d5ee01732g</a>
<b>Source type</b>	Research Article
<b>Keywords</b>	DACCS; techno-economic; life cycle; policy; public perception
<b>#Tags</b>	#DAC #Review #Policy
<b>Summary of key points + notes (include methodology)</b>	<p>Reviews over 800 publications on DAC and storage, using structured screening + ML classification</p> <p>Identifies main DAC categories (adsorbents, membranes, mineral looping, cryogenic, etc.)</p> <p>Discusses costs, lifecycle impacts, policy needs, MRV, and social acceptance</p>
<b>Research Question/Problem/ Need</b>	How can DACCS be scaled responsibly given high cost, energy draw, and social barriers?

<b>Important Figures</b>	
<b>VOCAB: (w/definition)</b>	<p>MRV — Monitoring, Reporting, Verification systems for CO<sub>2</sub> removal</p> <p>CDR — Carbon Dioxide Removal</p>
<b>Cited references to follow up on</b>	<p>Deutz, S., &amp; Bardow, A. (2021). Life-cycle assessment of an industrial direct air capture process ... <i>Nature Energy</i>, 6(1), 203–213. <a href="https://doi.org/10.1038/s41560-020-00771-9">https://doi.org/10.1038/s41560-020-00771-9</a></p> <p>Lück, S., et al. (2024). Machine-learning-assisted systematic review of DACCS literature. <i>Environmental Research Letters</i>, 19(4), 045002. <a href="https://doi.org/10.1088/1748-9326/accf82">https://doi.org/10.1088/1748-9326/accf82</a></p>
<b>Follow up Questions</b>	<p>What policy mechanisms (e.g. tax incentives) would best support gigaton-scale DAC deployment?</p> <p>How does public perception of DAC differ across global regions, and how should that affect siting?</p> <p>Which lifecycle uncertainties (energy source, sorbent production) are most critical to reduce?</p>

## Article #13 Notes: Electrochemical CO<sub>2</sub> capture with pH-independent redox chemistry

<b>Source Title</b>	Electrochemical CO <sub>2</sub> capture with pH-independent redox chemistry
<b>Source citation (APA Format)</b>	Kim, S. C., Gigantino, M., Holoubek, J., Matthews, J. E., Chen, J., Dho, Y., ... Tzeng, Y.-K. (2025). Electrochemical CO <sub>2</sub> capture with pH-independent redox chemistry. arXiv. <a href="https://arxiv.org/abs/2502.01028">https://arxiv.org/abs/2502.01028</a>

Original URL	<a href="https://arxiv.org/abs/2502.01028">https://arxiv.org/abs/2502.01028</a>
Source type	Experimental Article
Keywords	Redox capture; TEMPO; flow cell; low energy
#Tags	#Electrochemical #DAC #Redox
Summary of key points + notes (include methodology)	<p>Demonstrates a redox-based electrochemical DAC system using TEMPO in a flow cell</p> <p>Achieves ~2.6 kJ/mol CO<sub>2</sub> (very low energy) using pH-independent redox transition</p> <p>Uses computational modeling (MD + DFT) to rationalize molecular behavior</p>
Research Question/Problem/Need	How can electrochemical DAC operate at ultra-low energy costs using redox molecules?
Important Figures	 <p><b>Fig. 1.</b> Schematic diagram of the pH-independent of the electrochemical CO<sub>2</sub> capture system comprised of an electrochemical cell with symmetrical TEMPO redox reactions connected by a flow system to capture and release tanks.</p>

<b>VOCAB: (w/definition)</b>	Redox chemistry — electron-transfer reactions Flow cell — continuous electrochemical reactor TEMPO — a redox-active radical
<b>Cited references to follow up on</b>	Jaramillo, T. F., et al. (2021). Advances in electrochemical CO <sub>2</sub> capture technologies. ACS Energy Letters, 6(6), 1962–1977. <a href="https://doi.org/10.1021/acsenerylett.1c00648">https://doi.org/10.1021/acsenerylett.1c00648</a> Voskian, S., & Hatton, T. A. (2019). Faradaic electro-swing reactive adsorption for CO <sub>2</sub> capture. Energy & Environmental Science, 12(12), 3530–3547. <a href="https://doi.org/10.1039/C9EE02481C">https://doi.org/10.1039/C9EE02481C</a>
<b>Follow up Questions</b>	What is the long-term stability of TEMPO in cyclic operation? How could a scaled flow-cell system be designed cost-effectively? What renewable electricity sources could power such a system economically?

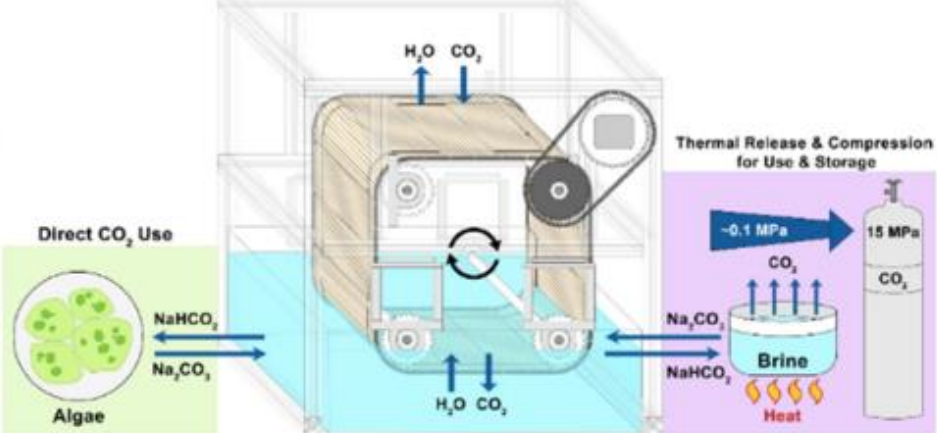
## Article #14 Notes: The Open DAC 2025 Dataset for Sorbent Discovery in Direct Air Capture

<b>Source Title</b>	The Open DAC 2025 Dataset for Sorbent Discovery in Direct Air Capture
<b>Source citation (APA Format)</b>	Sriram, A., Brabson, L. M., Yu, X., Choi, S., Abdelmaqsoud, K., Moubarak, E., ... Sholl, D. S. (2025). The Open DAC 2025 Dataset for Sorbent Discovery in Direct Air Capture. arXiv. <a href="https://arxiv.org/abs/2508.03162">https://arxiv.org/abs/2508.03162</a>
<b>Original URL</b>	<a href="https://arxiv.org/abs/2508.03162">https://arxiv.org/abs/2508.03162</a>
<b>Source type</b>	Dataset Article
<b>Keywords</b>	MOFs; DFT; machine learning; high-throughput; CO <sub>2</sub> sorbents
<b>#Tags</b>	#Materials #DAC #MOFs
<b>Summary of key points + notes (include methodology)</b>	Publishes a dataset of ~70 million DFT adsorption calculations (CO <sub>2</sub> , H <sub>2</sub> O, N <sub>2</sub> , O <sub>2</sub> ) over ~15,000 MOFs Provides ML-trained interatomic potentials for screening sorbent performance Designed to accelerate computational discovery of promising DAC materials
<b>Research Question/Problem/ Need</b>	How can high-throughput computational + ML approaches speed up the discovery of DAC sorbent materials?

<b>Important Figures</b>	<div style="display: flex; justify-content: space-around;"> <div style="text-align: center;"> <p><b>(a)</b></p> <p>— <math>N_2</math> (N = 29370) — <math>O_2</math> (N = 26296) — <math>CO_2</math> (N = 44883) — <math>H_2O</math> (N = 34634)</p> </div> <div style="text-align: center;"> <p><b>(b)</b></p> <table border="1"> <caption>Adsorption Energy Peak Distribution</caption> <thead> <tr> <th>Gas</th> <th>Percentage</th> </tr> </thead> <tbody> <tr> <td><math>H_2O</math></td> <td>78%</td> </tr> <tr> <td><math>CO_2</math></td> <td>13%</td> </tr> <tr> <td><math>O_2</math></td> <td>6%</td> </tr> <tr> <td><math>N_2</math></td> <td>1%</td> </tr> </tbody> </table> </div> </div>	Gas	Percentage	$H_2O$	78%	$CO_2$	13%	$O_2$	6%	$N_2$	1%
Gas	Percentage										
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$N_2$	1%										
<b>VOCAB: (w/definition)</b>	<p>DFT — density functional theory (quantum-level calculation)</p> <p>MOF — metal-organic framework</p> <p>Henry's law coefficient — gas uptake per partial pressure</p>										
<b>Cited references to follow up on</b>	<p>Sriram, A., et al. (2024). ODAC23 dataset: earlier version for MOFs in DAC screening. ACS Central Science, 10(6), 923–935.  <a href="https://doi.org/10.1021/acscentsci.4c00323">https://doi.org/10.1021/acscentsci.4c00323</a></p>										
<b>Follow up Questions</b>	<p>Which MOFs in ODAC25 show favorable adsorption under humid, real-DAC conditions?</p> <p>How quickly can these top computational candidates be validated experimentally?</p> <p>What are the limitations of ML potentials, and how can future models be improved?</p>										

## Article #15 Notes: Design and demonstration of a direct air capture system with moisture-driven $CO_2$ delivery into aqueous medium

<b>Source Title</b>	Design and demonstration of a direct air capture system with moisture-driven $CO_2$ delivery into aqueous medium
<b>Source citation (APA Format)</b>	Flory, J., Taylor, S., Li, S., Tiwari, S., Cole, G., Lowe, A., ... Vermaas, W. (2025). Design and demonstration of a direct air capture system with moisture-driven $CO_2$ delivery into aqueous medium. arXiv. <a href="https://arxiv.org/abs/2508.02650">https://arxiv.org/abs/2508.02650</a>
<b>Original URL</b>	<a href="https://arxiv.org/abs/2508.02650">https://arxiv.org/abs/2508.02650</a>

<b>Source type</b>	Experimental Article
<b>Keywords</b>	Moisture swing; ion-exchange resin; DAC; water evaporation
<b>#Tags</b>	#DAC #Sorbent #MoistureSwing
<b>Summary of key points + notes (include methodology)</b>	<p>Built a DAC prototype using resin beads to absorb CO<sub>2</sub> via moisture swing</p> <p>Outdoor and lab demonstration, achieving ~100 g/day in pilot</p> <p>Techno-economic model: ~\$670/t capture; aspirational future: ~\$51/t</p>
<b>Research Question/Problem/ Need</b>	Can moisture-swing resin systems achieve DAC at low energy and water cost?
<b>Important Figures</b>	 <p><b>Graphical Abstract.</b> Schematic of moisture-driven direct air capture system with CO<sub>2</sub> delivery into aqueous medium for direct use to cultivate algae or thermal extraction and compression for use, conversion or storage.</p>
<b>VOCAB: (w/definition)</b>	<p>Ion-exchange resin — polymer that exchanges ions with environment</p> <p>Moisture swing — using humidity to drive adsorption/desorption</p>
<b>Cited references to follow up on</b>	Lackner, K. S., et al. (2012). Carbon dioxide extraction from air: is the economics realistic? <i>Energy</i> , 45(1), 110–119. <a href="https://doi.org/10.1016/j.energy.2012.03.057">https://doi.org/10.1016/j.energy.2012.03.057</a>
<b>Follow up Questions</b>	<p>How can water use be reduced in a moisture-swing DAC system?</p> <p>What are the degradation mechanisms for resin beads under repeated humidity cycles?</p> <p>Can this technology be powered by intermittent renewables to minimize cost?</p>

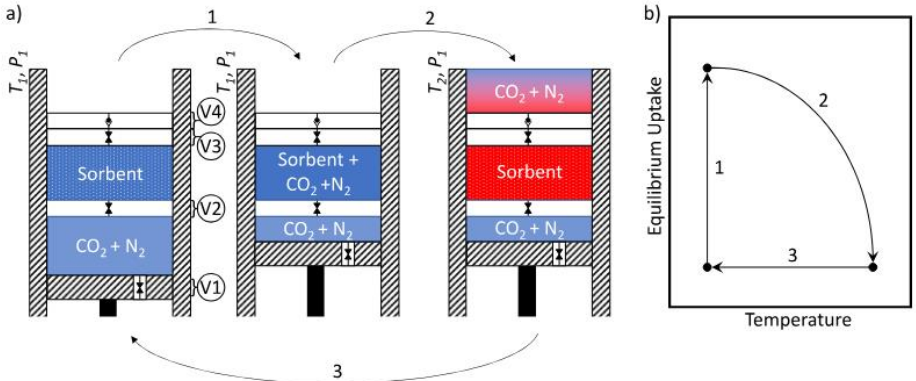
# Article #16 Notes: Enhancing Direct Air Capture through Potassium Carbonate Doping of Activated Carbons

<b>Source Title</b>	Enhancing Direct Air Capture through Potassium Carbonate Doping of Activated Carbons
<b>Source citation (APA Format)</b>	van Dongen, N., van Hoof, A. J. F., Calero, S., & Vicent-Luna, J. M. (2025). Enhancing Direct Air Capture through Potassium Carbonate Doping of Activated Carbons. arXiv. <a href="https://arxiv.org/abs/2510.06400">https://arxiv.org/abs/2510.06400</a>
<b>Original URL</b>	<a href="https://arxiv.org/abs/2510.06400">https://arxiv.org/abs/2510.06400</a>
<b>Source type</b>	Computational Article
<b>Keywords</b>	Activated carbon; $K_2CO_3$ ; adsorption; $CO_2$ ; humidity
<b>#Tags</b>	#Adsorbents #DAC #CarbonMaterials
<b>Summary of key points + notes (include methodology)</b>	<p>Molecular simulations show <math>K_2CO_3</math>-doped carbon has better <math>CO_2</math> and water adsorption than undoped; <math>K_2CO_3</math> clusters provide strong binding sites</p> <p>Doped pores favor <math>CO_2</math> adsorption at low partial pressure, which is beneficial for DAC</p>
<b>Research Question/Problem/ Need</b>	Can doping activated carbon with $K_2CO_3$ improve its effectiveness for DAC in humid air?
<b>Important Figures</b>	<div style="text-align: right;">7</div> <p>The figure consists of six subplots arranged in a 2x3 grid, labeled (a) through (f). Each plot shows the adsorption loading (in mol/kg) as a function of either partial pressure (Pa) or relative humidity (%). The top row (a, b, c) shows <math>CO_2</math> adsorption, and the bottom row (d, e, f) shows <math>H_2O</math> adsorption. The left column (a, d) compares 'no <math>K_2CO_3</math>' (red), '5 <math>K_2CO_3</math> random' (orange), and '10 <math>K_2CO_3</math> random' (yellow). The middle column (b, e) compares 'no <math>K_2CO_3</math>' (red), '5 <math>K_2CO_3</math> clusters' (orange), and '10 <math>K_2CO_3</math> clusters' (yellow). The right column (c, f) compares '5 <math>K_2CO_3</math> random' (orange), '5 <math>K_2CO_3</math> clusters' (orange), '10 <math>K_2CO_3</math> random' (yellow), and '10 <math>K_2CO_3</math> clusters' (yellow). Plots (c) and (f) include a vertical dashed line at approximately <math>10^2</math> Pa and <math>60\%</math> RH, respectively, indicating a region of interest for DAC. In general, the 'clusters' show higher adsorption capacity than 'random' distributions, and the '10 <math>K_2CO_3</math>' conditions show higher capacity than the '5 <math>K_2CO_3</math>' conditions.</p>
<b>VOCAB: (w/definition)</b>	<p>Doping — adding a chemical additive to a material to modify its properties</p> <p>Adsorption isotherm — equilibrium uptake vs pressure</p>

<b>Cited references to follow up on</b>	McPherson, L. R., & Rood, M. J. (1993). Adsorption of CO <sub>2</sub> on activated carbon at low partial pressure. <i>Carbon</i> , 31(4), 599–605. <a href="https://doi.org/10.1016/0008-6223(93)90079-H">https://doi.org/10.1016/0008-6223(93)90079-H</a>
<b>Follow up Questions</b>	<p>What is the thermal or pressure energy required to regenerate the doped carbon?</p> <p>How stable is the K<sub>2</sub>CO<sub>3</sub> in the carbon matrix during many adsorption/desorption cycles?</p> <p>Can experimental samples be produced matching the doped structures from simulation?</p>

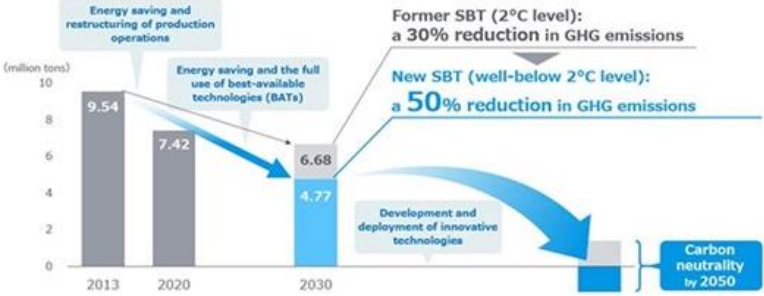
## Article #17 Notes: Intrinsic Direct Air Capture

<b>Source Title</b>	Intrinsic Direct Air Capture
<b>Source citation (APA Format)</b>	McDannald, A., Siderius, D. W., DeCost, B., Choudhary, K., Ortiz-Montalvo, D. L., & ... (2025). Intrinsic Direct Air Capture. arXiv. <a href="https://arxiv.org/abs/2501.04825">https://arxiv.org/abs/2501.04825</a>
<b>Original URL</b>	<a href="https://arxiv.org/abs/2501.04825">https://arxiv.org/abs/2501.04825</a>
<b>Source type</b>	Theoretical Article
<b>Keywords</b>	Intrinsic metrics; MOFs; thermodynamics; performance bounds
<b>#Tags</b>	#Theory #MOFs #MaterialScreening
<b>Summary of key points + notes (include methodology)</b>	<p>Defines thermodynamic “intrinsic” metrics for sorbents that are independent of cycle design</p> <p>Applies these metrics to ~11,660 MOFs using equilibrium uptake data</p> <p>Finds that the path through pressure/temperature cycle space strongly influences performance</p>
<b>Research Question/Problem/ Need</b>	How can intrinsic thermodynamic limits improve materials screening for DAC?

<b>Important Figures</b>	<p><b>Model</b></p> 
<b>VOCAB: (w/definition)</b>	<p>Equilibrium uptake — amount of gas held by a sorbent at a given temperature and pressure, at equilibrium</p> <p>Intrinsic metric — a performance measure independent of a specific DAC system design</p>
<b>Cited references to follow up on</b>	<p>DeCost, B., &amp; Sholl, D. S. (2023). A thermodynamic screening metric for CO<sub>2</sub> capture performance of sorbents. <i>The Journal of Physical Chemistry C</i>, 127(10), 4750–4760. <a href="https://doi.org/10.1021/acs.jpcc.2c06534">https://doi.org/10.1021/acs.jpcc.2c06534</a></p>
<b>Follow up Questions</b>	<p>Which MOFs approach the intrinsic performance limits defined in this work?</p> <p>How do real DAC cycle performances compare to these intrinsic predictions?</p> <p>Can new MOFs be designed explicitly to maximize these intrinsic metrics?</p>

## Article #18 Notes: Carbon capture and recycling technology of carbon resources under the target of carbon neutrality

<b>Source Title</b>	<p>Carbon capture and recycling technology of carbon resources under the target of carbon neutrality</p>
<b>Source citation (APA Format)</b>	<p>Al-Mujtaba, K., Haddad, M., &amp; ... (2024). Carbon capture and recycling technology of carbon resources under the target of carbon neutrality. <i>International Journal of Low-Carbon Technologies</i>, 19(6), 2693–2701. <a href="https://academic.oup.com/ijlct/article/doi/10.1093/ijlct/ctae229/7901309">https://academic.oup.com/ijlct/article/doi/10.1093/ijlct/ctae229/7901309</a></p>
<b>Original URL</b>	<p><a href="https://academic.oup.com/ijlct/article/doi/10.1093/ijlct/ctae229/7901309">https://academic.oup.com/ijlct/article/doi/10.1093/ijlct/ctae229/7901309</a></p>

<b>Source type</b>	Research Article
<b>Keywords</b>	CCU; methanation; hydrogen; CO <sub>2</sub> recycling
<b>#Tags</b>	#CCU #Methanation #Hydrogen
<b>Summary of key points + notes (include methodology)</b>	Reviews technologies to recycle CO <sub>2</sub> into synthetic fuels using green hydrogen Emphasizes potential CO <sub>2</sub> capture efficiency >90% and integration with existing gas infrastructure Discusses economic and logistic hurdles (H <sub>2</sub> production, transport, emissions)
<b>Research Question/Problem/ Need</b>	How can CCU (capture + recycling) be made economically and environmentally viable using renewable hydrogen?
<b>Important Figures</b>	 <p><b>Figure 3</b> Magnificent plan to reach carbon neutrality-to achieve carbon neutrality, CCS, or CCUS, is essential.</p>
<b>VOCAB: (w/definition)</b>	Methanation — CO <sub>2</sub> + H <sub>2</sub> → CH <sub>4</sub> reaction Carbon recycling — using CO <sub>2</sub> as a feedstock rather than simply storing it
<b>Cited references to follow up on</b>	Aramco, S., & Linde, B. (2023). CO <sub>2</sub> utilization for synthetic methane: process design and techno-economic analysis. <i>Journal of CO<sub>2</sub> Utilization</i> , 69, 102798. <a href="https://doi.org/10.1016/j.jcou.2023.102798">https://doi.org/10.1016/j.jcou.2023.102798</a>
<b>Follow up Questions</b>	What scale of green hydrogen production is needed to support CCU at gigaton CO <sub>2</sub> /year? How do lifecycle emissions compare for CCU vs CCS? What business models make CCU financially viable, considering infrastructure and policy?

## Article #19 Notes: A comprehensive review of enhanced CO<sub>2</sub> capture using activated carbon derived from biomass feedstock

<b>Source Title</b>	A comprehensive review of enhanced CO <sub>2</sub> capture using activated carbon derived from biomass feedstock
<b>Source citation (APA Format)</b>	Kundu, S., Khandaker, T., Anik, M. A. M., Hasan, M. K., Dhar, P. K., Dutta, S. K., Latif, M. A., & Hossain, M. S. (2024). A comprehensive review of enhanced CO <sub>2</sub> capture using activated carbon derived from biomass feedstock. <i>RSC Advances</i> , 14(40), 29693–29736. <a href="https://pubs.rsc.org/en/Content/ArticleLanding/2024/RA/D4RA04537H">https://pubs.rsc.org/en/Content/ArticleLanding/2024/RA/D4RA04537H</a>
<b>Original URL</b>	<a href="https://pubs.rsc.org/en/Content/ArticleLanding/2024/RA/D4RA04537H">https://pubs.rsc.org/en/Content/ArticleLanding/2024/RA/D4RA04537H</a>
<b>Source type</b>	Review Article
<b>Keywords</b>	Biomass; activated carbon; CO <sub>2</sub> adsorption; sustainability
<b>#Tags</b>	#Biomass #Adsorbents #SustainableCCS
<b>Summary of key points + notes (include methodology)</b>	<p>Surveys biomass-derived activated carbons made from agricultural residues</p> <p>Compares activation methods, porosity, capacity, regeneration energy</p> <p>Highlights sustainability potential and cost advantages, but variability in feedstock and performance remain a challenge</p>
<b>Research Question/Problem/ Need</b>	Can biomass-derived activated carbon deliver cost-effective, sustainable CO <sub>2</sub> sorbents at scale?
<b>Important Figures</b>	<p>The diagram illustrates the thermal decomposition of biomass components across different temperature ranges. A horizontal arrow at the bottom indicates temperature from 100°C to 600°C. Above it, several boxes describe processes:</p> <ul style="list-style-type: none"> <li><b>Moisture drying</b> (0-100°C)</li> <li><b>Hemicellulose decomposition</b> (200-350°C)</li> <li><b>Cellulose decomposition</b> (250-400°C)</li> <li><b>Lignin</b> (250-530°C)</li> <li><b>200-320°C - Torrefaction</b> giving better fuels for combustion and gasification applications</li> <li><b>300-600°C - Fast giving bio-oil</b> that can be converted to bio-diesel</li> <li><b>&gt; 600°C Carbonization</b> for biochar and activated porous carbons</li> </ul>

<b>VOCAB: (w/definition)</b>	Activated carbon — porous carbon made (often by pyrolysis + activation) for adsorption  Biomass feedstock — organic waste used as raw material
<b>Cited references to follow up on</b>	Sevilla, M., & Fuertes, A. B. (2009). The production of carbon materials by hydrothermal carbonization of cellulose. <i>Carbon</i> , 47(9), 2281–2289. <a href="https://doi.org/10.1016/j.carbon.2009.04.035">https://doi.org/10.1016/j.carbon.2009.04.035</a>
<b>Follow up Questions</b>	What is the performance variability across carbons from different biomass sources?  How does the energy/carbon cost of producing biomass carbon compare to synthetic sorbents?  How many cycles (adsorption/desorption) can biomass-based carbon survive before performance drops?

## Article #20 Notes: Revolutionizing carbon capture efficiency: A comprehensive review of AI-Driven approaches

<b>Source Title</b>	Revolutionizing carbon capture efficiency: A comprehensive review of AI-Driven approaches
<b>Source citation (APA Format)</b>	Anonymous authors. (2025). Revolutionizing carbon capture efficiency: A comprehensive review of AI-driven approaches. <i>International Journal of Science, Technology &amp; Research Applications</i> , 9(1), 43–68. <a href="https://sciresjournals.com/ijstra/sites/default/files/IJSTRA-2025-0043.pdf">https://sciresjournals.com/ijstra/sites/default/files/IJSTRA-2025-0043.pdf</a>
<b>Original URL</b>	<a href="https://sciresjournals.com/ijstra/sites/default/files/IJSTRA-2025-0043.pdf">https://sciresjournals.com/ijstra/sites/default/files/IJSTRA-2025-0043.pdf</a>
<b>Source type</b>	Review Article
<b>Keywords</b>	Machine learning; process optimization; sorbent discovery; monitoring; CCUS
<b>#Tags</b>	#AI #CCUS #Optimization
<b>Summary of key points + notes (include methodology)</b>	Reviews AI applications in CCUS: ML for materials, DRL for process control, anomaly detection for storage  Reports use cases where cost was cut ~10–20% when AI optimized parameters

	Suggests that AI can reduce R&D time, improve sorbent lifetime, and support risk monitoring
<b>Research Question/Problem/Need</b>	How can AI methods accelerate materials discovery and process scale-up for CCUS while reducing costs and risk?
<b>Important Figures</b>	<p>The diagram, titled "Role of AI &amp; Hybrid Modeling", illustrates a process flow. On the left, a database icon labeled "Data" has a green arrow pointing into a blue funnel. On the right, an icon of three people with a lightbulb labeled "Human Expertise" has an orange arrow pointing into the same funnel. Inside the funnel, the text "AI + Simulation" is displayed above icons of a brain and a computer monitor. A thick blue arrow points downwards from the bottom of the funnel to the text "Enriched Knowledge for Well-Informed Decision Making". Below this, a white box with a blue border contains the text: "Optimized value chain &amp; well-informed decisions to accelerate CCUS global adoption".</p>
<b>VOCAB: (w/definition)</b>	<p>Deep Reinforcement Learning — AI technique where learning agents optimize decisions over time</p> <p>Machine Learning — predictive modelling based on patterns in data</p>
<b>Cited references to follow up on</b>	Gómez-Bombarelli, R., et al. (2018). Design of efficient molecular organic light-emitting diodes by a high-throughput virtual screening ... <i>Nature</i> , 562(7728), 86–90. <a href="https://doi.org/10.1038/s41586-018-0626-6">https://doi.org/10.1038/s41586-018-0626-6</a>
<b>Follow up Questions</b>	<p>What datasets are most needed to build robust AI models for CCUS?</p> <p>How reliable are AI-predicted lifetimes of sorbents versus real-world cycling data?</p>

	Can reinforcement learning optimize the full CCUS chain (capture → transport → storage)?
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## Article #21 Notes (Patent #1): Air purification and disinfection apparatus and methods of use

<b>Source Title</b>	Air purification and disinfection apparatus and methods of use
<b>Source citation (APA Format)</b>	Pisharodi, M. (2022, November 29). <i>US11511013B2 - air purification and disinfection apparatus and methods of use</i> . Google Patents. <a href="https://patents.google.com/patent/US11511013B2/en?q=%28Carbon%2BDioxide%2BFilter%29&amp;oq=Carbon%2BDioxide%2BFilter">https://patents.google.com/patent/US11511013B2/en?q=%28Carbon%2BDioxide%2BFilter%29&amp;oq=Carbon%2BDioxide%2BFilter</a>
<b>Original URL</b>	<a href="https://patents.google.com/patent/US11511013B2/en?q=(Carbon+Dioxide+Filter)&amp;oq=Carbon+Dioxide+Filter">https://patents.google.com/patent/US11511013B2/en?q=(Carbon+Dioxide+Filter)&amp;oq=Carbon+Dioxide+Filter</a>
<b>Source type</b>	Patent
<b>Keywords</b>	Carbon Dioxide Purification Portable Contaminants Airborne
<b>#Tags</b>	
<b>Summary of key points + notes (include methodology)</b>	<p>Personal Air Purification &amp; Disinfection: The invention focuses on a portable system that purifies and disinfects ambient air for an individual user, e.g., a healthcare worker or person in a confined space.</p> <p>Multi-Stage Air Treatment: Air enters a housing where it may pass through HEPA filtration, activated carbon filters, and UV-C disinfection chambers to remove particulates, microorganisms, chemical vapors, and other contaminants.</p> <p>Ultraviolet Disinfection: UV-C and Far-UV light sources within multiple chambers kill/neutralize airborne microorganisms by exposing them to ultraviolet radiation with extended residence times through serpentine/helical air paths.</p> <p>Carbon Dioxide &amp; Gas Absorption: The apparatus can include replaceable carbon dioxide absorption units to reduce CO<sub>2</sub> in the breathed air when required (e.g., in closed circuits).</p>

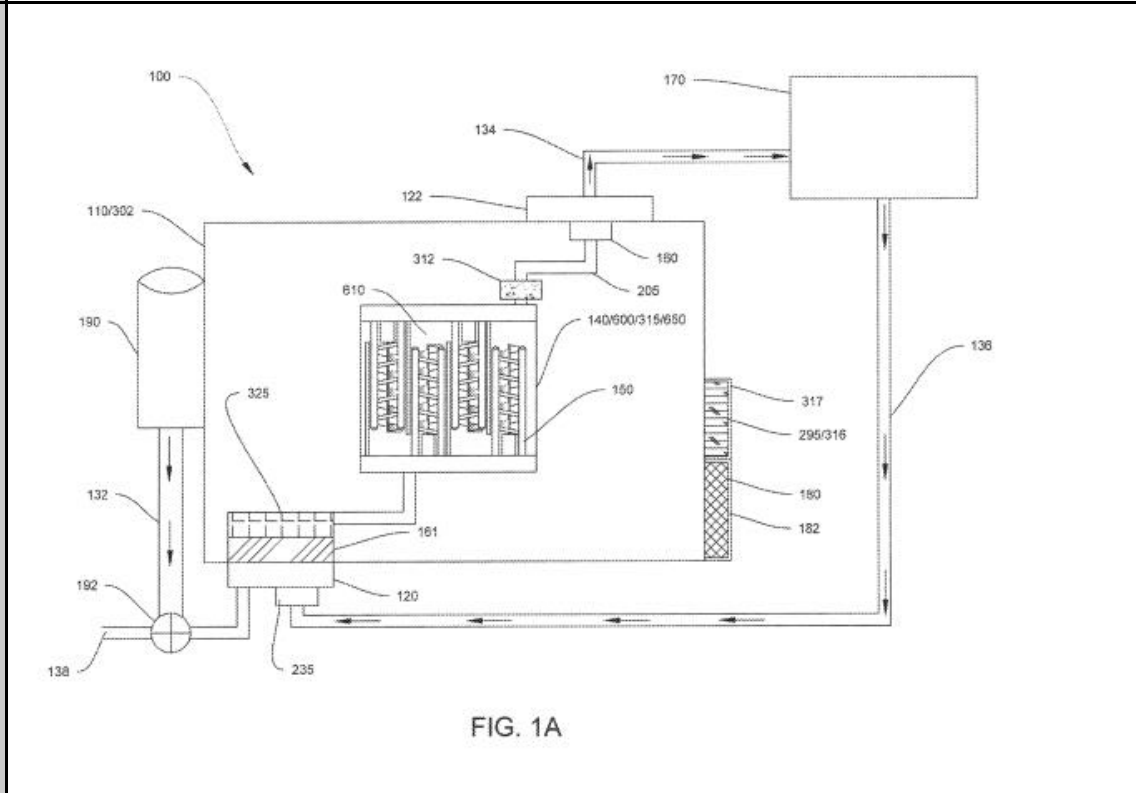
**Active Airflow Control:** The system uses an air mover (pump or fan) and a pump speed controller (PCB) to regulate airflow through filters, UV chambers, and to the user's distribution unit (mask/ventilator).

**Modular / Wearable Configurations:** The device can take many physical forms: backpack, briefcase, wearable cervical collar, or integrated respirator-style unit. This accommodates diverse use-case scenarios.

**Closed and Open Circuit Modes:** Embodiments include systems that either release exhaled air back into the unit for purification (closed-circuit) or discharge it externally (open-circuit), providing flexibility depending on application needs

**Research Question/Problem/ Need** How can a portable air purification and disinfection apparatus be designed to effectively remove airborne particulates, pathogens and other contaminants to deliver purified air directly to a user?

**Important Figures**



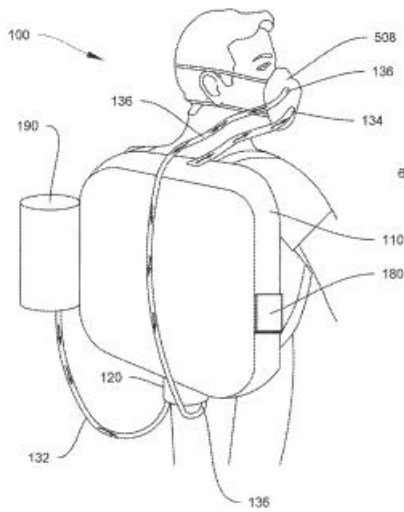


FIG. 1B

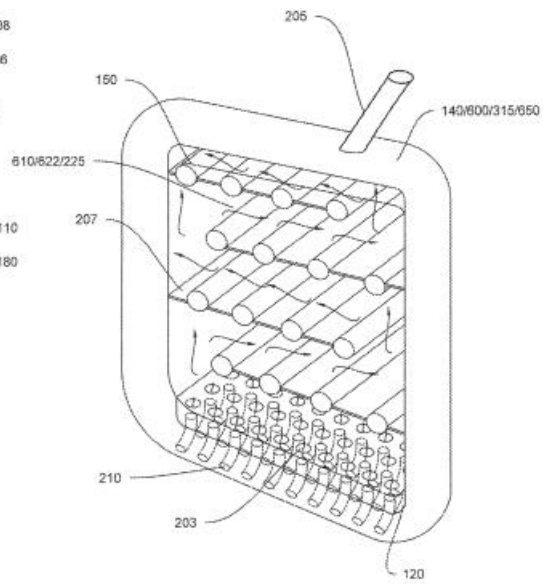


FIG. 2A

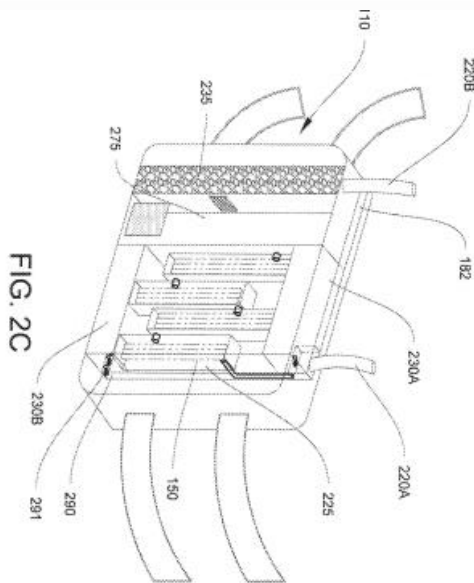


FIG. 2C

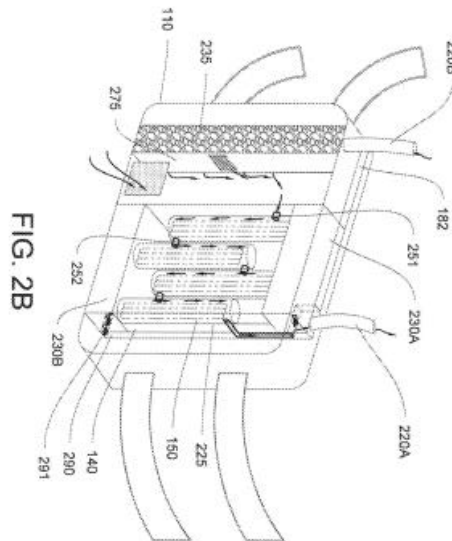


FIG. 2B

**VOCAB:  
(w/definition)**

HEPA – A type of air filter that removes at least 99.97% of airborne particles that are 0.3 microns in size, such as dust, pollen, mold, and bacteria.

PCB - A flat board that holds and connects electronic components and allows electricity to flow between them in a controlled way.

Apparatus - A device or system of connected parts designed to perform a specific function or task, often used in technical or scientific contexts.

<b>Cited references to follow up on</b>	Document entitled: 3818 U1 Mobile Filter for Respirator Masks With UVC Sources and Electronic Control Unit, machine translation of BG 3818 provided by Google Translate (Year: 2020).* How a packaged system works (Goodman) Jul. 29, 2016, [online] retrieved from <URL: <a href="https://web.archive.org/web/20160729193422/https://www.goodmanmfg.com/resources/heating-cooling-101/how-a-packaged-system-works">https://web.archive.org/web/20160729193422/https://www.goodmanmfg.com/resources/heating-cooling-101/how-a-packaged-system-works</a> >.
<b>Follow up Questions</b>	<p>What are the specific performance benefits and limitations of including a carbon dioxide absorption unit within this system?</p> <p>How does the system balance airflow rate, UV exposure time, and filter/absorbent media durability to ensure both effective microbial kill rates and user comfort?</p> <p>What are the design trade-offs for integrating multiple treatment technologies in a single portable housing?</p>

## Article #22 Notes (Patent #2): Purifying carbon dioxide using activated carbon

<b>Source Title</b>	Purifying carbon dioxide using activated carbon
<b>Source citation (APA Format)</b>	Degenstein, N., Shah, M., & Neu, B. (2012, October 9). <i>US8282715B1 - purifying carbon dioxide using activated carbon</i> . Google Patents. <a href="https://patents.google.com/patent/US8282715B1/en?q=%28Carbon%2BDioxide%2BFilter%29&amp;oq=Carbon%2BDioxide%2BFilter">https://patents.google.com/patent/US8282715B1/en?q=%28Carbon%2BDioxide%2BFilter%29&amp;oq=Carbon%2BDioxide%2BFilter</a>
<b>Original URL</b>	<a href="https://patents.google.com/patent/US8282715B1/en?q=(Carbon+Dioxide+Filter)&amp;oq=Carbon+Dioxide+Filter">https://patents.google.com/patent/US8282715B1/en?q=(Carbon+Dioxide+Filter)&amp;oq=Carbon+Dioxide+Filter</a>
<b>Source type</b>	Patent
<b>Keywords</b>	Adsorption/Adsorbent Flue Gas Carbon Amines
<b>#Tags</b>	
<b>Summary of key points + notes (include methodology)</b>	<p>Activated carbon adsorption is used to purify CO<sub>2</sub>-rich gas streams by selectively removing contaminants such as SO<sub>x</sub>, NO<sub>x</sub>, and trace impurities.</p> <p>The invention targets industrial exhaust and flue gas, enabling cleaner carbon dioxide suitable for sequestration, reuse, or further processing.</p>

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Chemical interactions on the carbon surface convert sulfur and nitrogen oxides into adsorbed species, improving removal efficiency beyond simple physical filtration.

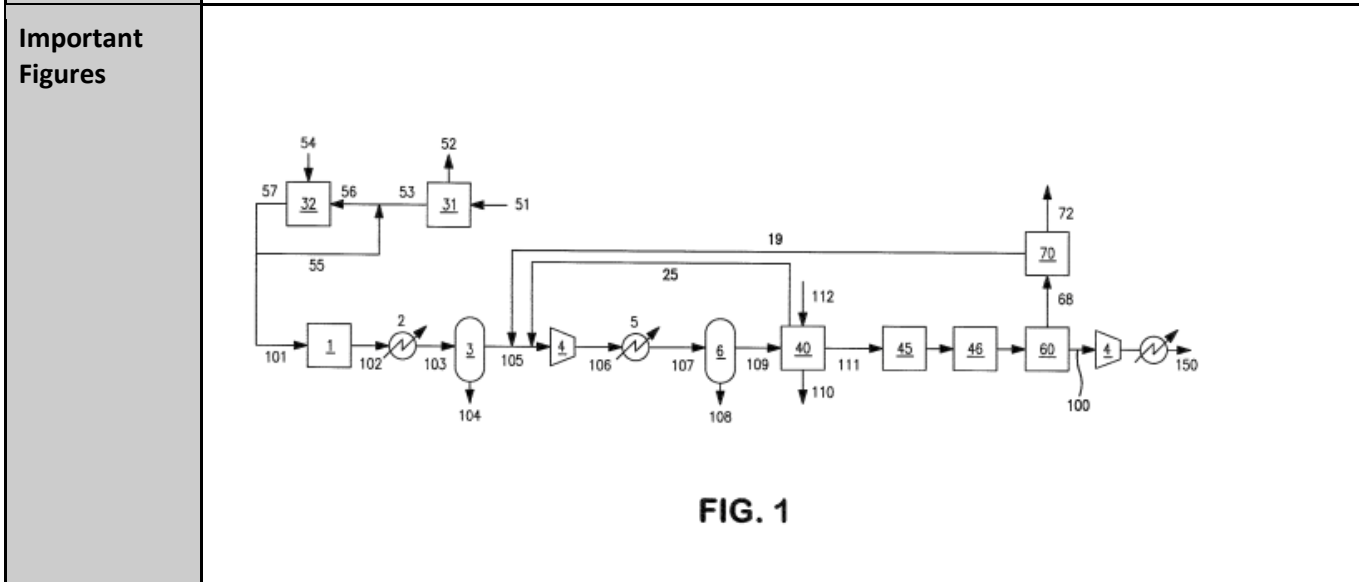
The system includes regenerable adsorbent beds, allowing activated carbon to be reused rather than replaced after saturation.

Washing or regeneration steps (often using water) remove adsorbed acidic compounds from the carbon, restoring adsorption capacity.

The process may operate using pressure-swing adsorption (PSA) or vacuum PSA, improving control over adsorption and desorption cycles.

Overall, the patent claims a cost-effective, scalable alternative to solvent-based CO<sub>2</sub> purification methods (such as amines), with lower corrosion and operational complexity.

**Research Question/Problem/ Need** How can an impure carbon dioxide-rich gas stream be effectively purified using regenerable activated carbon adsorption techniques to produce higher-purity CO<sub>2</sub> for reuse, sequestration, or industrial applications?



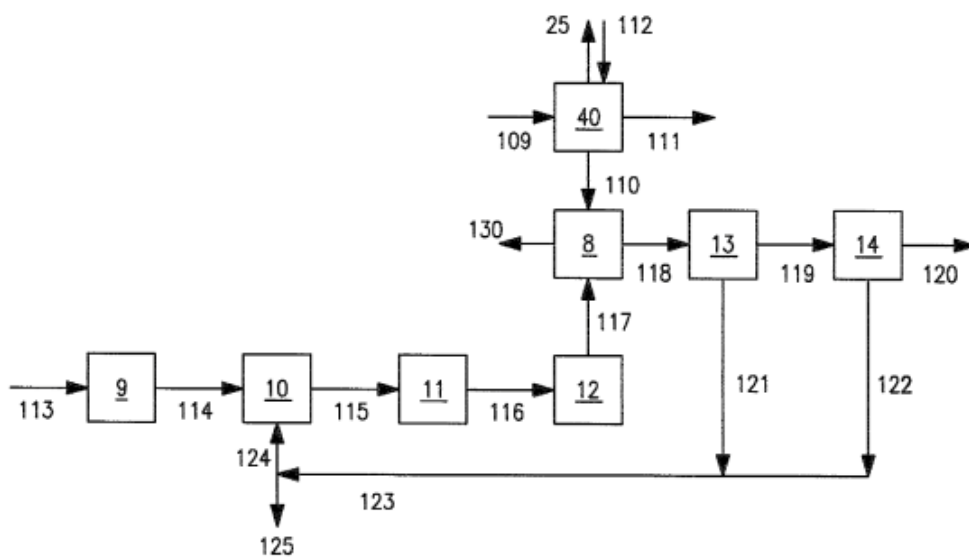


FIG. 2

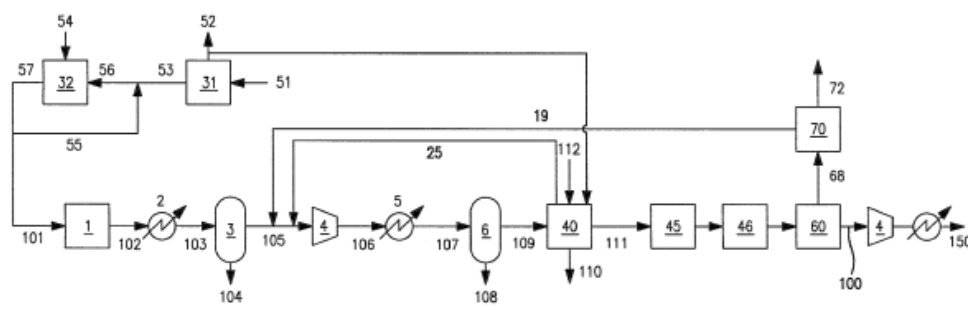


FIG. 3

**VOCAB:  
(w/definition  
)**

Activated Carbon Adsorption – method to purify CO<sub>2</sub>-rich gas streams by selectively removing contaminants

Pressure Swing Adsorption - improves control over adsorption and desorption cycles

**Cited  
references to  
follow up on**

Metzinger J, et al., "Application of a Periodically Operated Trickle Bed to Sulfur Removal from Stack Gas." *Chemical Engineering Science*, (47) 3723-3727, 1992.

Metzinger J, et al., "A Novel Periodic Reactor for Scrubbing SO<sub>2</sub> from Industrial Stack Gas." *Chemical Engineering Science*, (49) 4533-4546, 1994.

	Vladea R, et al., "High-Efficiency Structured-Packing Catalysts with Activated Carbon for SO <sub>2</sub> Oxidation from Flue Gas" Energy & Fuels, (11) 277-283, 1997.
<b>Follow up Questions</b>	<p>What are the specific performance advantages of using the patented activated carbon system compared with traditional solvent-based CO<sub>2</sub> purification methods like amine scrubbing?</p> <p>How does the regeneration process for the activated carbon beds affect operational downtime and energy consumption?</p> <p>What types and concentrations of contaminants can the system tolerate before performance degrades, and how does that influence the choice of carbon material?</p>